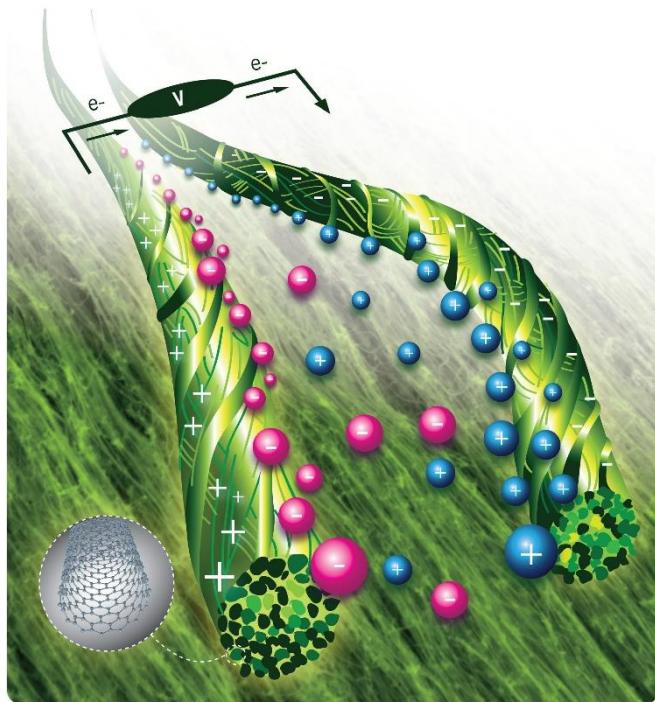


Aligeramiento de vehículos y aeronaves mediante nanomateriales



Materiales para una mejor defensa
TALLER ORGANIZADO POR LA ACADEMIA DE
LAS CIENCIAS Y ARTES MILITARES (ACAMI)
E IMDEA MATERIALES

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5 diciembre 2024

Dr Juan José Vilatela

Investigador en IMDEA Materiales

<https://www.materials.imdea.org/groups/mng/>
[@MNGMaterials](https://twitter.com/MNGMaterials)

Encargado de proyectos de I+D industriales del sector transporte (aviónica, automoción, aeronáutica) con empresas reconocidas del sector (Toyota, Airbus, RockwellCollins, Shell)

Proyectos financiados por el ejército y fuerza aérea de EEUU

CTO de [Floatech](#), empresa fabricante de electrodos nanoestructurados

Profesor asociado:

UC3M, Máster Materiales, Grado Ing. (Ciencia de Materiales, Física)
UPM, Máster Materiales, Máster Compuestos Airbus



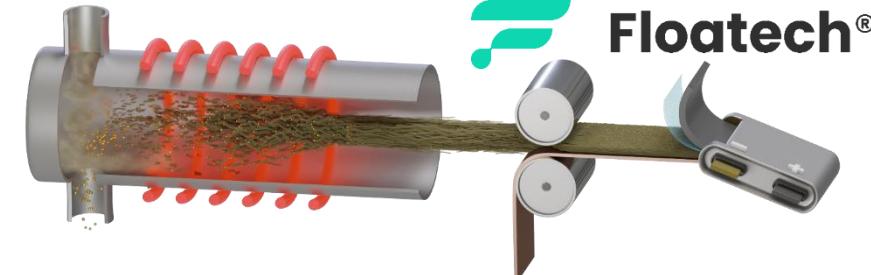
Discover Nano

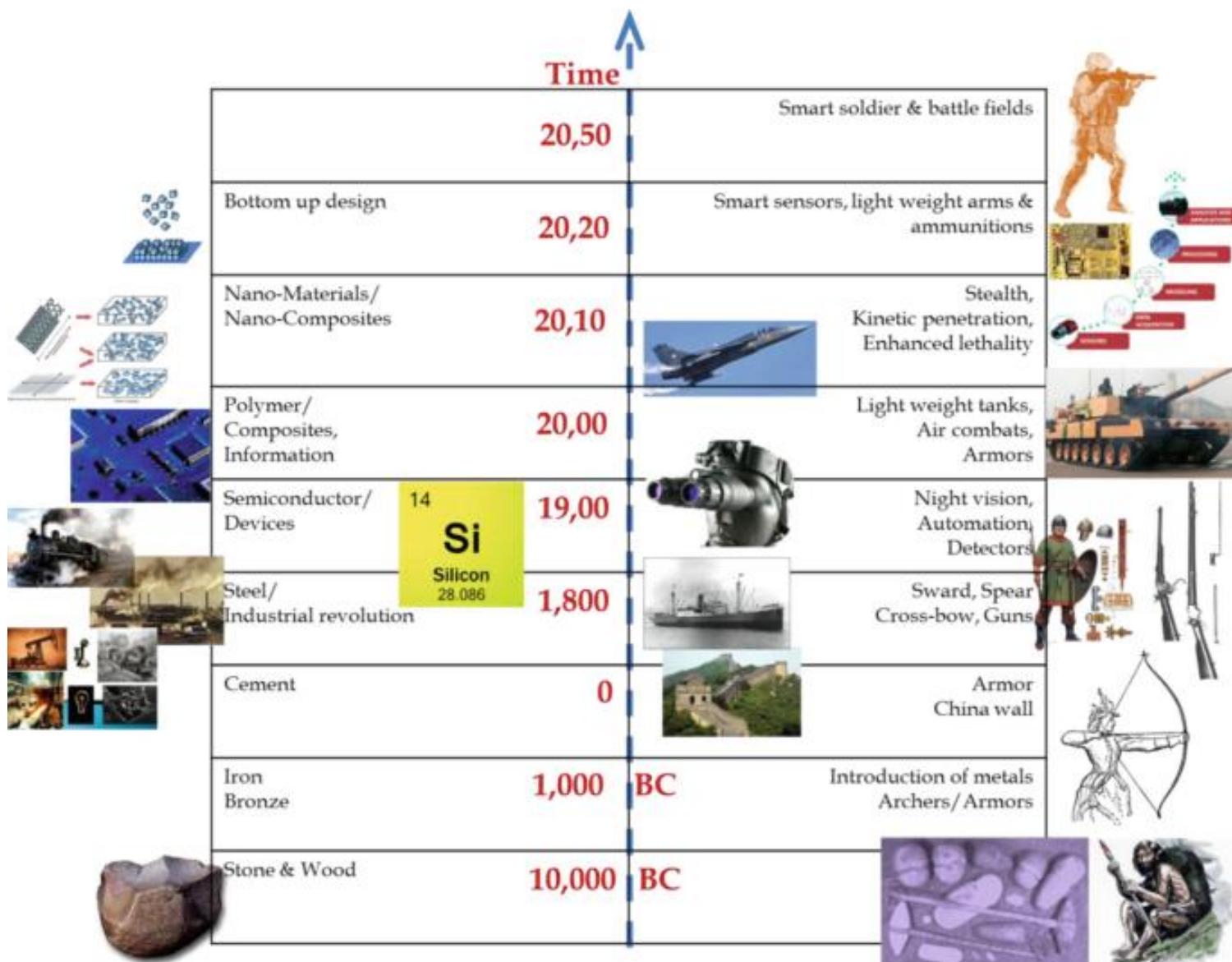
Review

Opportunities for nanomaterials in more sustainable aviation

Afshin Pendashteh¹ · Anastasija Mikhalchan¹ · Tamara Blanco Varela² · Juan J. Vilatela¹

Received: 6 February 2024 / Accepted: 21 August 2024





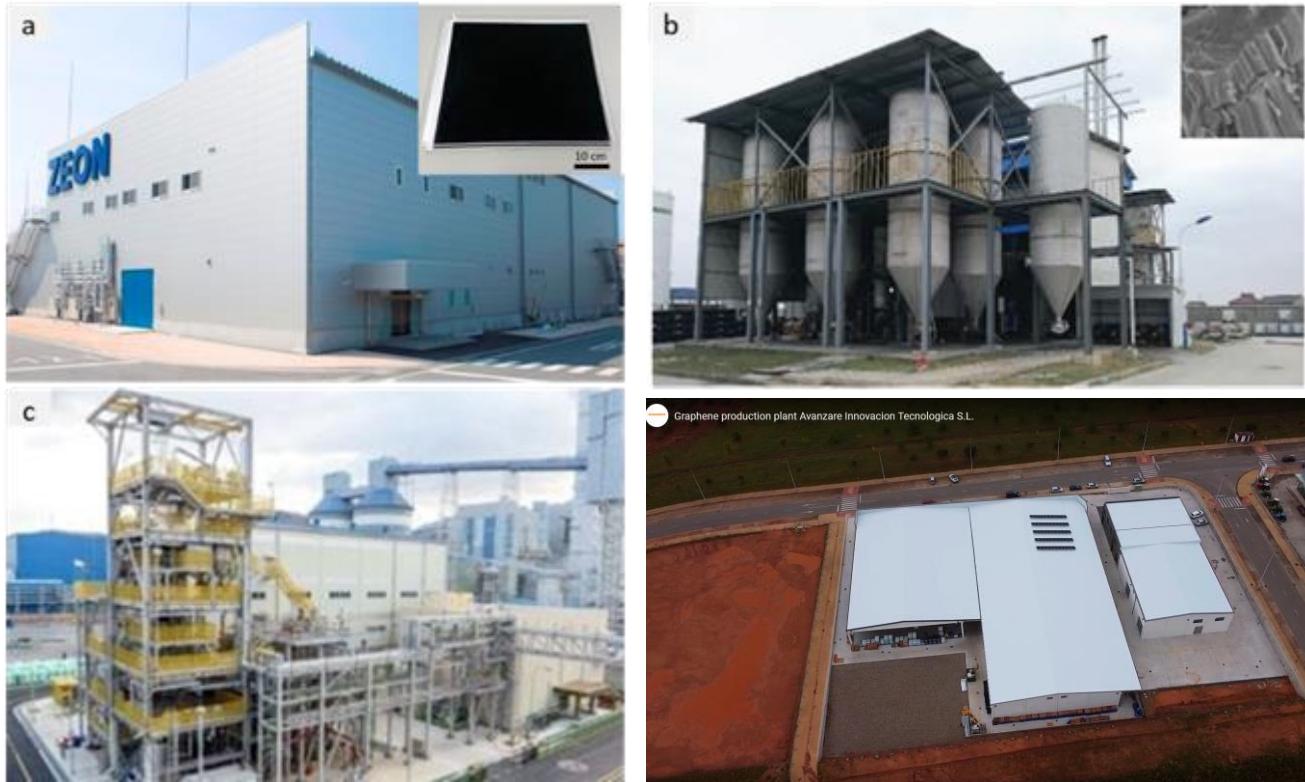


Aligeramiento a través de:

- 1) Baterías de nueva generación
- 2) Conductores eléctricos supermetálicos
- 3) Materiales estructurales para balística
- 4) Apantallamiento electromagnético y emisores de campo para guerra electrónica
- 5) Sensores químicos textiles

¿Cuál es el más importante de los nanomateriales usados en la actualidad?

Examples of industrial facilities for nanomaterials production



a) Zeon Corporation plant in Japan; b) 1000 tons per year plant facilities of CNT arrays developed by Tsinghua University (China) [19]; c) LG Chem's Yeosu Plant (South Korea) for 2000 ton per reactor d) Graphene producer Avanzare.

Annual production is 10 000 tons per year, growing at about 20% CAGR

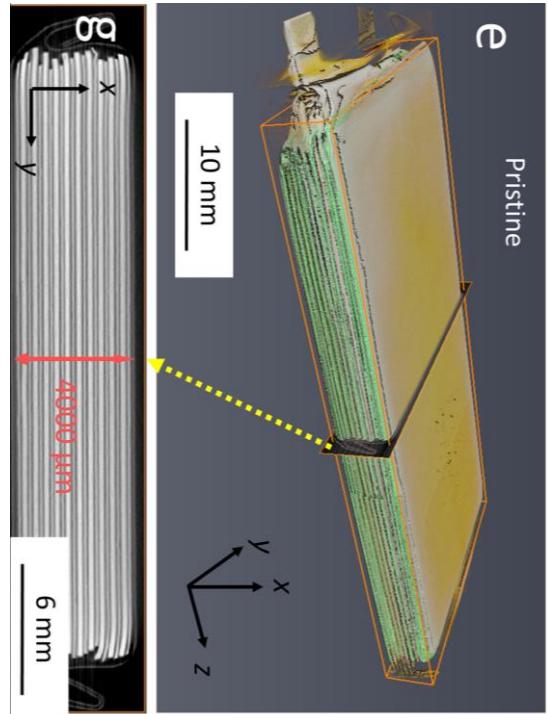
¿Dónde se usan?

Background: battery electrode and cell assembly

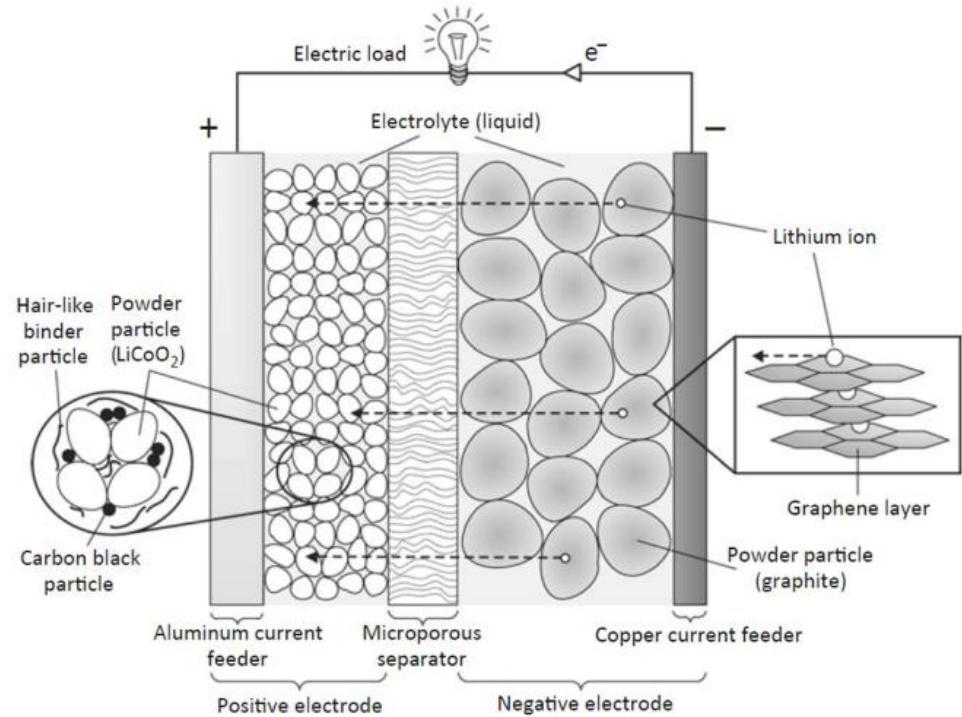
Battery



Multi-electrode construction

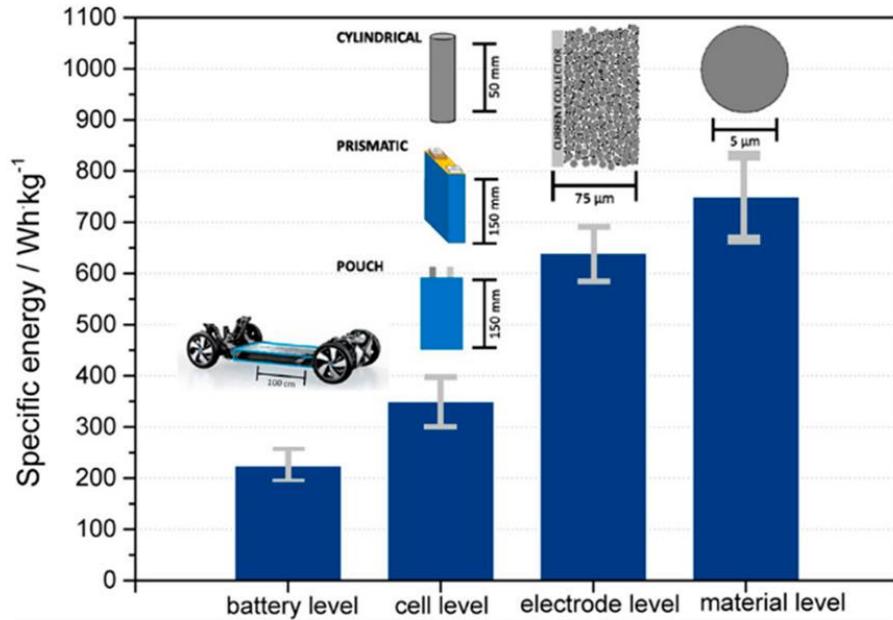


Scheme of a single cell unit



Component breakdown in LIB cell

Specific energy targets from material to pack level

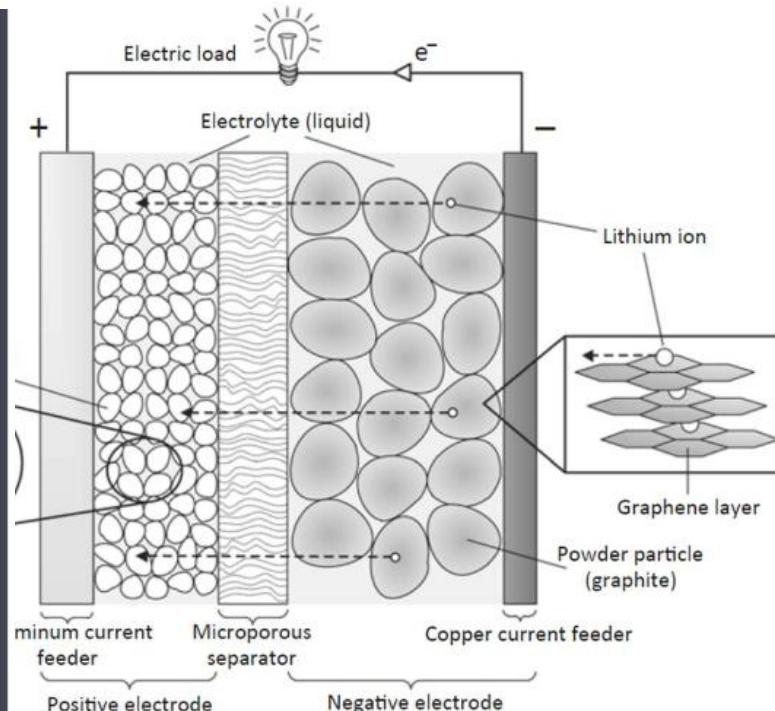
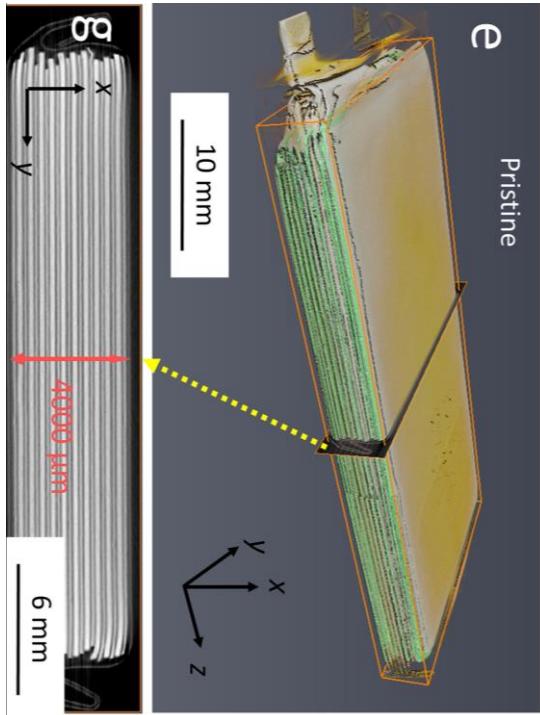


	Specific energy, Wh/kg
Material level	750
Electrode level	650
Cell level	350
Battery level	210

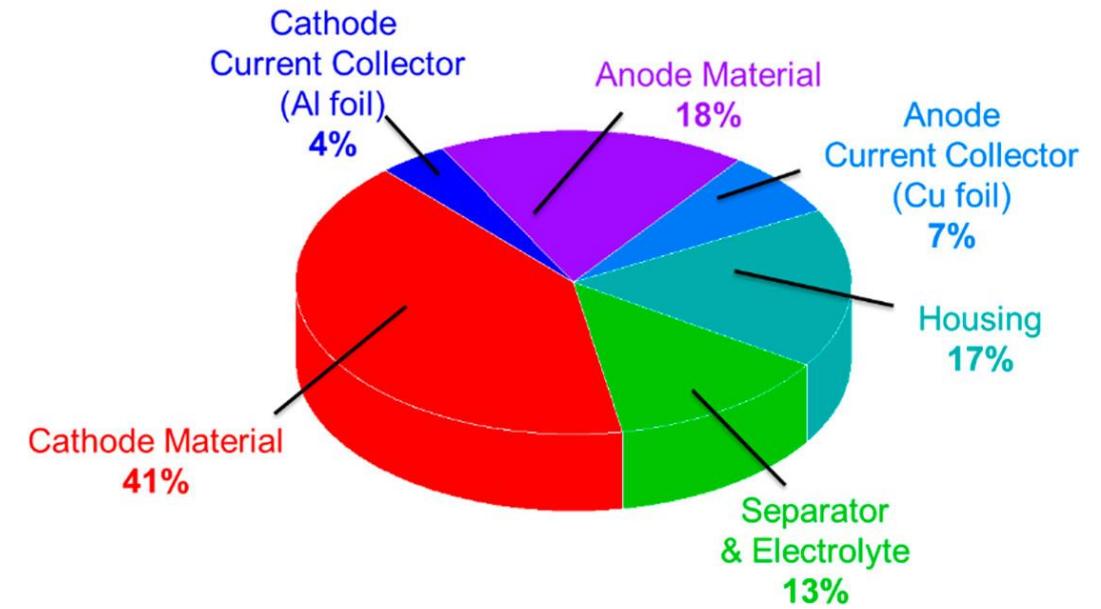
(b) Specific energy targets for automotive applications from material to pack level.

What are the limitations to reduce the weight of auxiliary elements in the cell?

Component breakdown in LIB cell

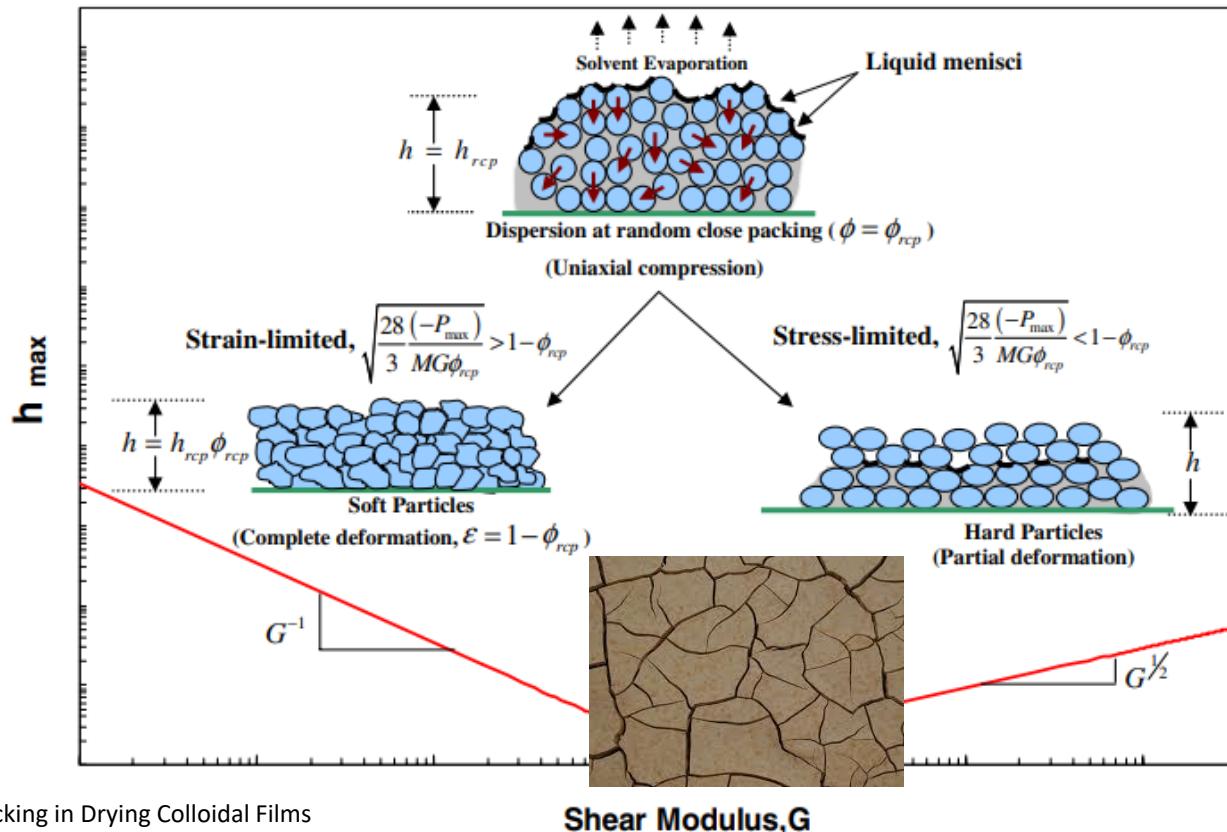


Weight occupied by Li-ion cell components



What are the limitations to reduce the weight of auxiliary elements?

Limitations on electrode thickness



Cracking in Drying Colloidal Films
Karnail B. Singh and Mahesh S. Tirumkudulu
Phys. Rev. Lett. 98, 218302 –2007

$$\text{Critical crack thickness (CCT), } h_{max} = 0.41 \sqrt{\frac{GMV_f R^3}{2\gamma}}$$

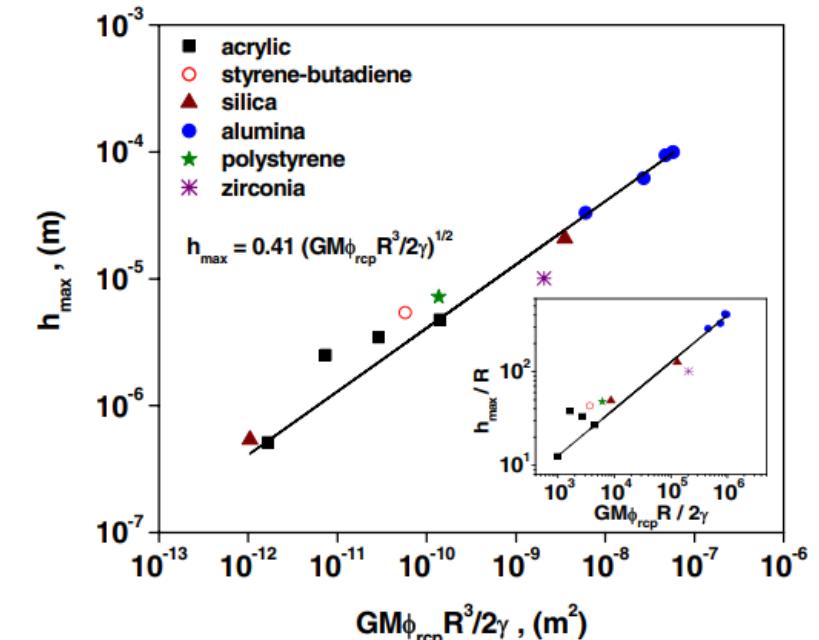
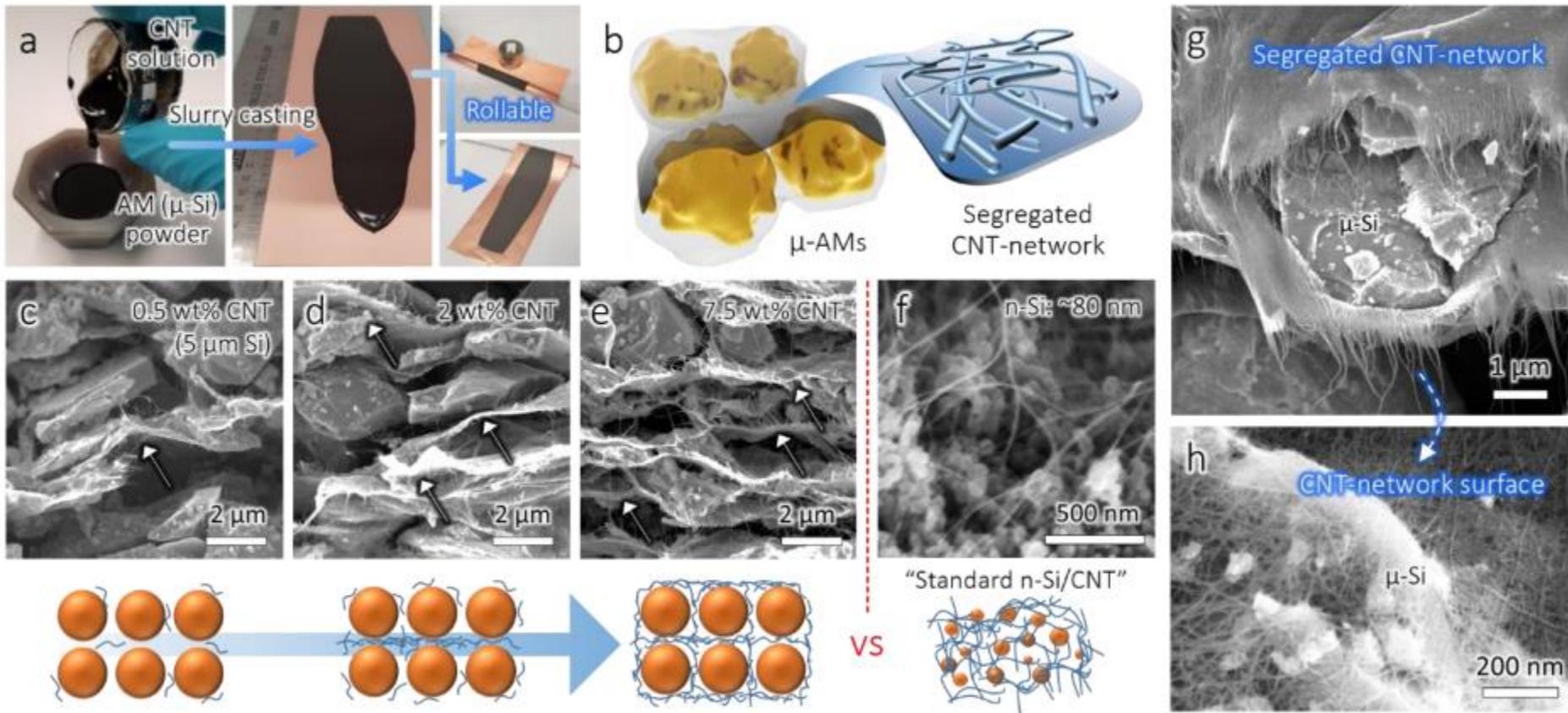
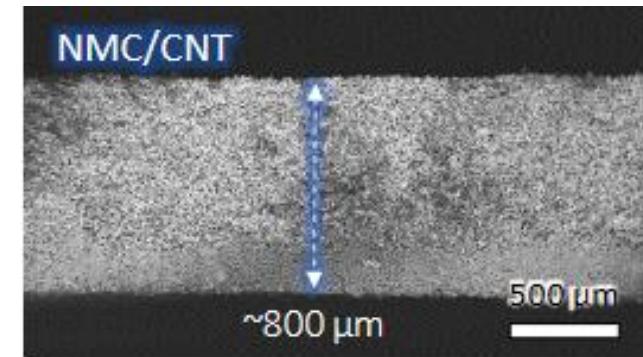
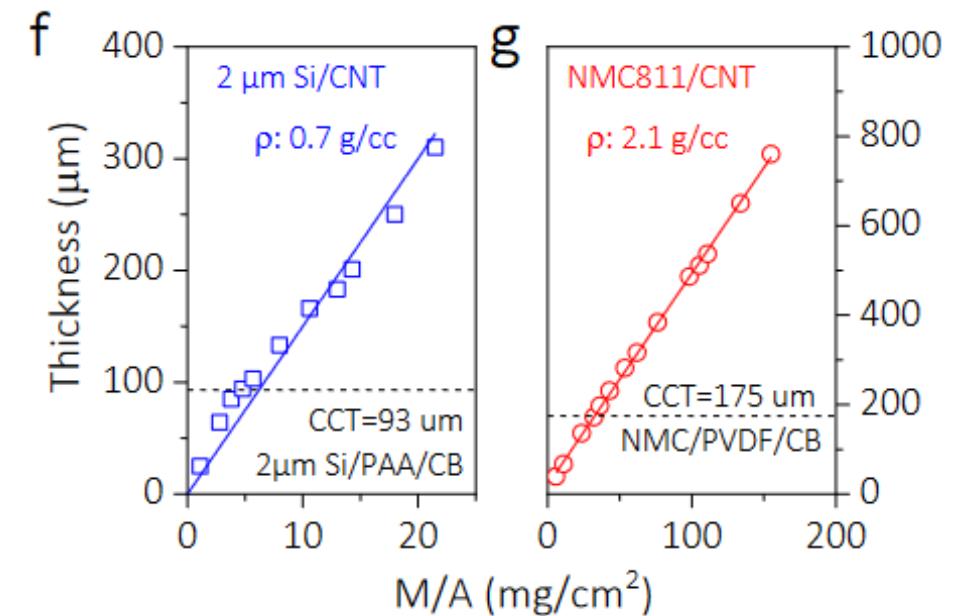
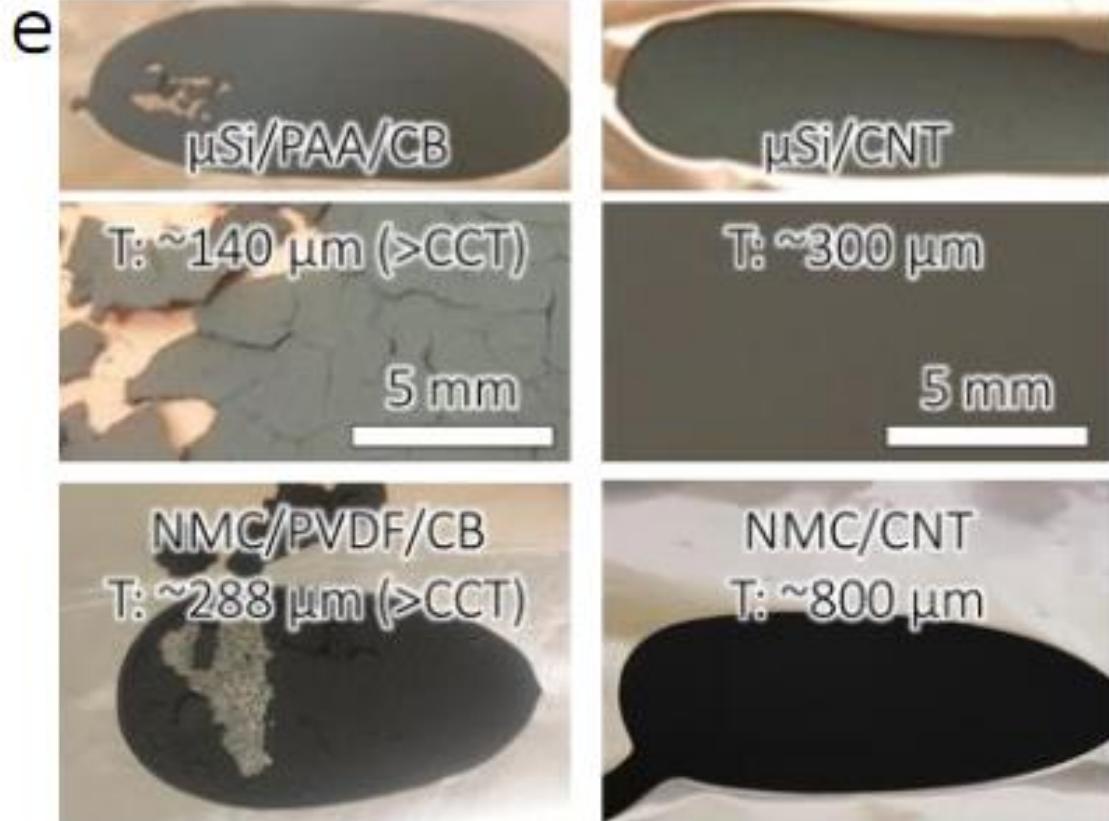


FIG. 3 (color online). The measured critical thickness vs the characteristic scale, $\frac{GM\phi_{rcp}R^3}{2\gamma}$. The data points are for films of acrylic [■, $G = 0.8$ GPa, $2R = 82, 133, 206,$ and 353 nm, and $\phi_{rcp}(M) = 0.65(6.7), 0.66(6.8), 0.68(7.0),$ and $0.67(6.9)$, respectively], styrene-butadiene [○, $G = 1$ GPa, $2R = 250$ nm, and $\phi_{rcp}(M) = 0.64(6.6)$], silica [▲, $G = 31$ GPa, $2R = 330,$ and 22 nm, and $\phi_{rcp}(M) = 0.60(6.1)$ for both], alumina [●, $G = 156$ GPa, $2R = 230, 379, 458,$ and 489 nm, and $\phi_{rcp}(M) = 0.60(6.1)$ for all], polystyrene [★, $G = 1.6$ GPa, $2R = 300$ nm, and $\phi_{rcp}(M) = 0.60(6.1)$], and zirconia [*, $G = 81$ GPa, $2R = 200$ nm, and $\phi_{rcp}(M) = 0.60(6.1)$]. Here the value of surface tension, γ , is taken as 0.072 N/m for all the cases. The solid line is a power law with an exponent $1/2$ and the multiplying coefficient is obtained via regression. The inset plots the same data in the nondimensional form.

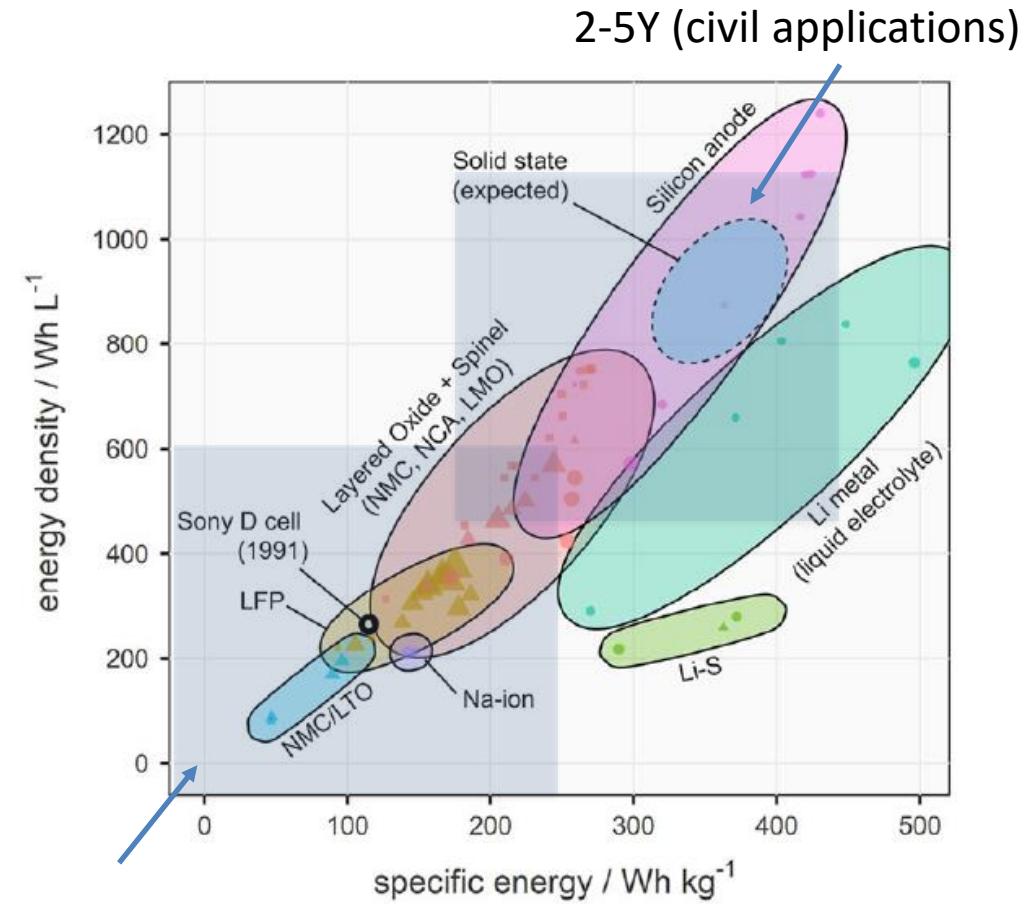
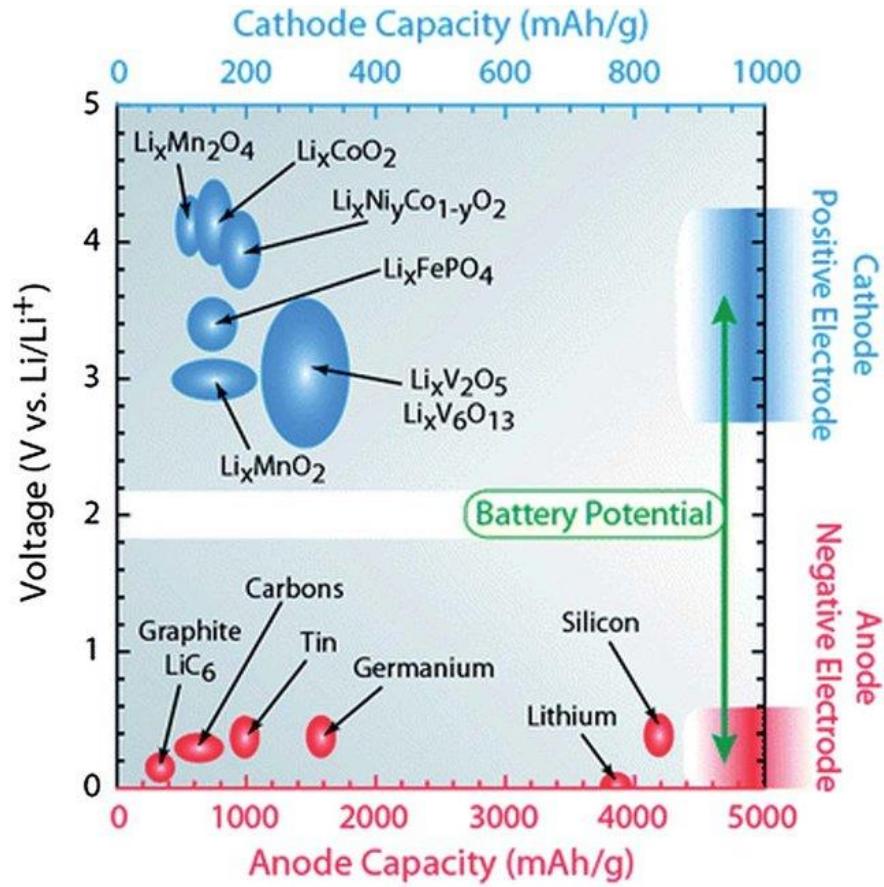
Limitations on electrode thickness



Limitations on electrode thickness

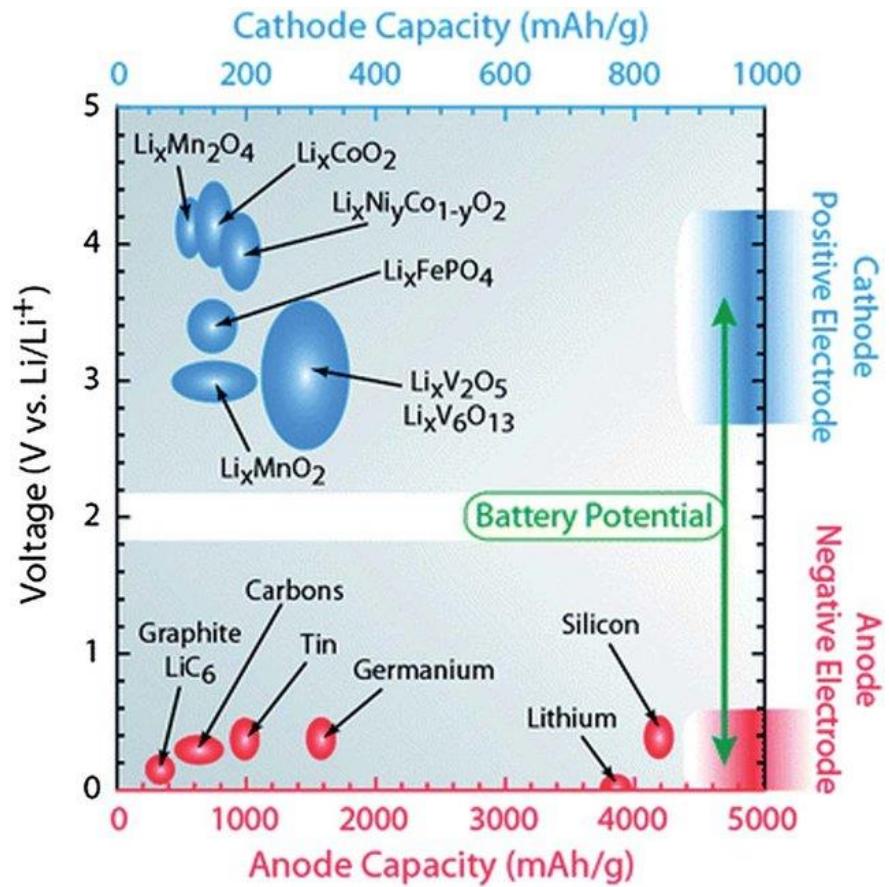


New generation of batteries



Today

Background: battery basics



$$E = V \times Q$$

$$V_{cell} = V_{cathode} - V_{anode}$$

Consider an LFP cathode-graphite anode battery:

$$V_{cell} = 3.55 - (0.05) = 3.5V$$

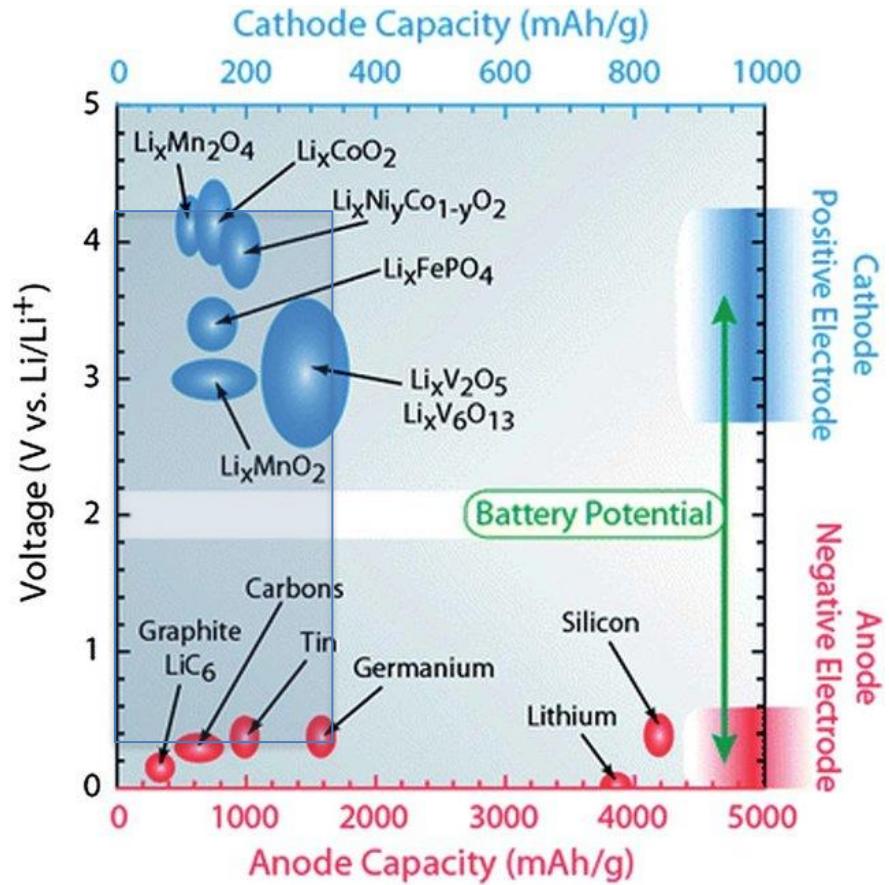
$$Q/m = 300 \text{ mAh/g}$$

Rough estimate of energy density

$$E_s \approx \frac{1}{2} * \frac{Q}{m} * V_{cell} = 350 \text{ Wh/kg}$$

Compare to today's SoA automotive EV Battery (250 Wh/kg)

Background: battery basics



$E = V \times Q$, but Q is a function of charge rate

$$V_{cell} = V_{cathode} - V_{anode}$$

Consider an LFP cathode-graphite anode battery:

$$V_{cell} = 3.55 - (0.05) = 3.0V$$

$$Q/m = 300 \text{ mAh/g}$$

Rough estimate of energy density

$$E_s \approx \frac{1}{2} * \frac{Q}{m} * V_{cell} = 300 \text{ Wh/kg}$$

Compare to today's SoA automotive EV Battery (250 Wh/kg)

Background: battery basics

Two main sources of limitations in energy density relative to our theoretical estimate:
various resistances and device constraints.

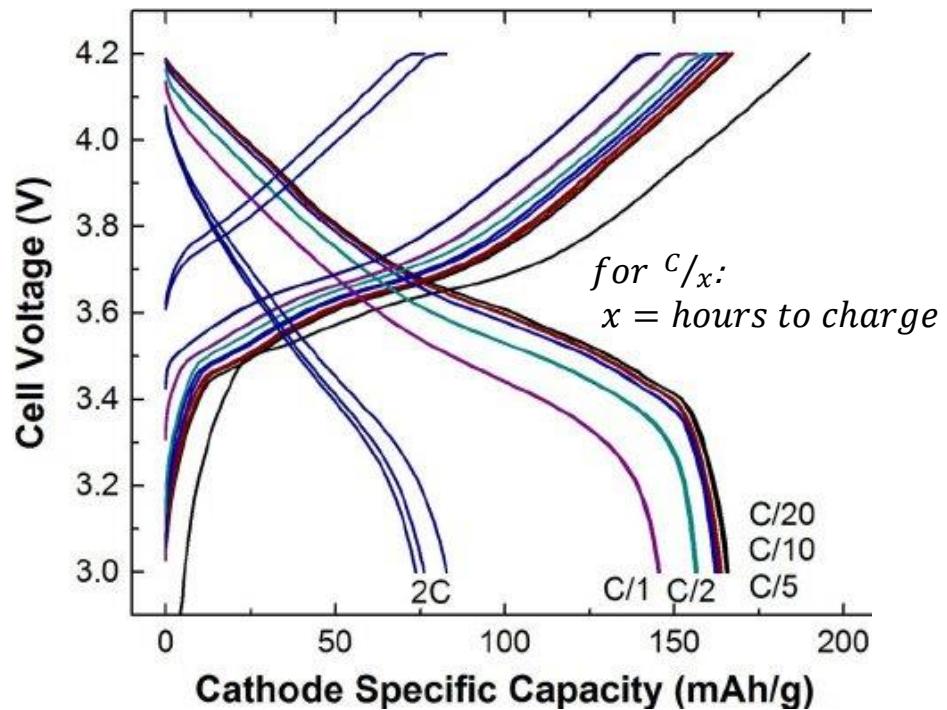


Figure 9. Initial voltage profiles for a 45 mAh NMC622/Gr pouch cell at C/20, C/10, C/5, C/2, 1C, and 2C discharge rate between 3 and 4.2 V utilizing electrodes with 3.3 mAh/cm^2 available areal capacities. The charge rates were symmetric to discharge until a maximum of C/3 rate. A voltage hold trickle charged the cell at 4.2 V until the measured current was less than that corresponding to C/20.

Capacity is a function of current density

$$Q_{\text{electrode}} = Q_{\text{electrode}}(J)$$

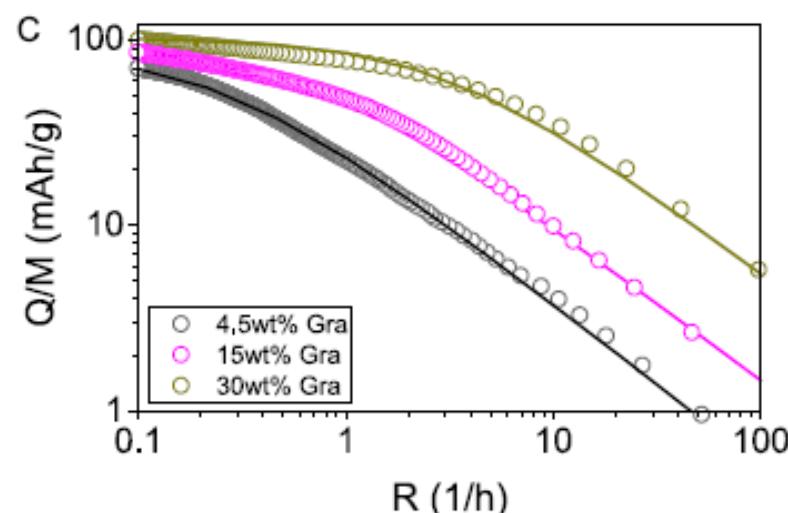
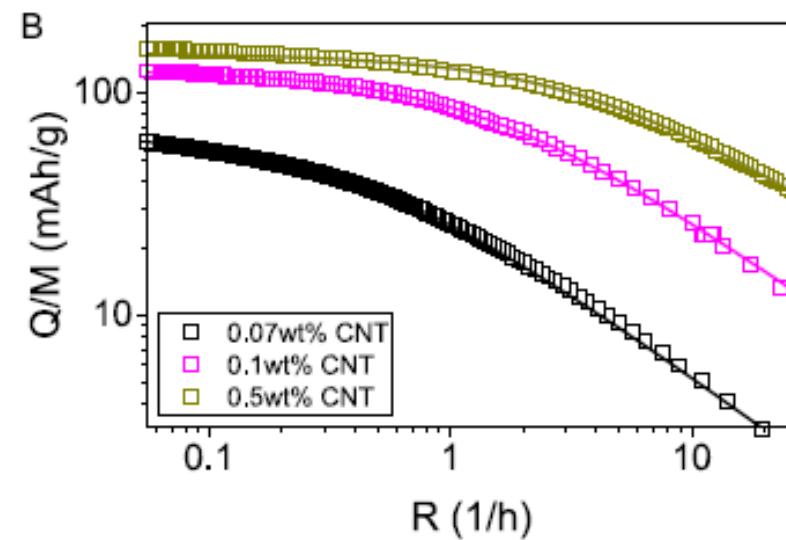
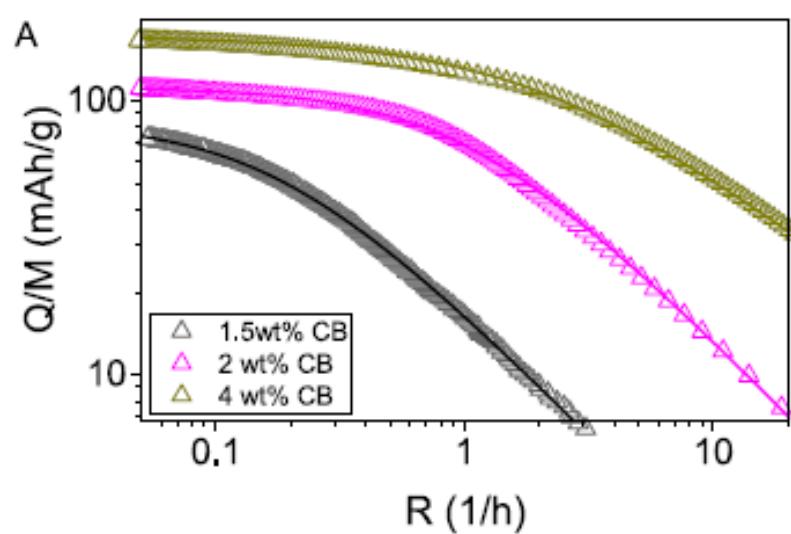
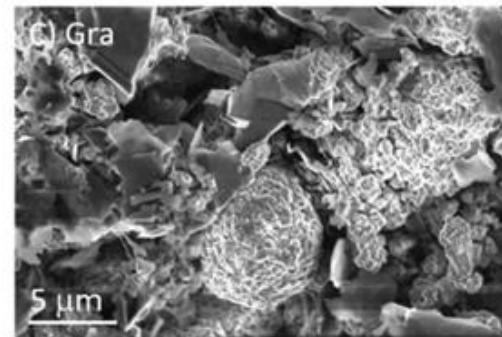
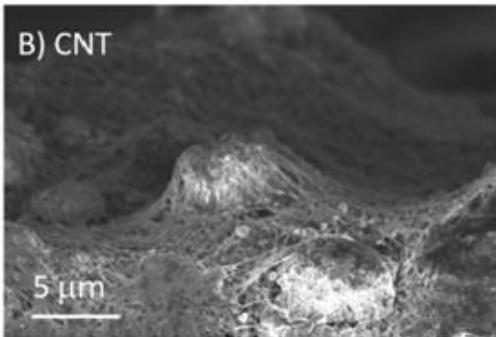
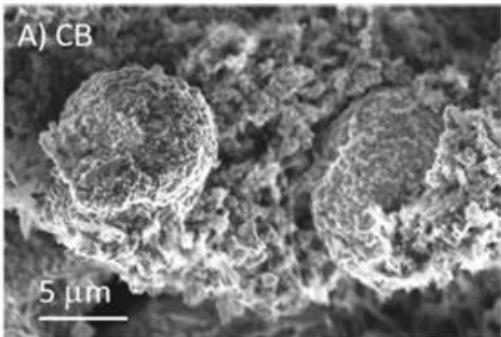
The real energy density normalised by active materials is

$$E_{s,am} \approx \frac{1}{2} * \frac{1}{m} \int_0^Q V_{cell} * dQ$$

Consider the case of 2C (charging in 30 min):

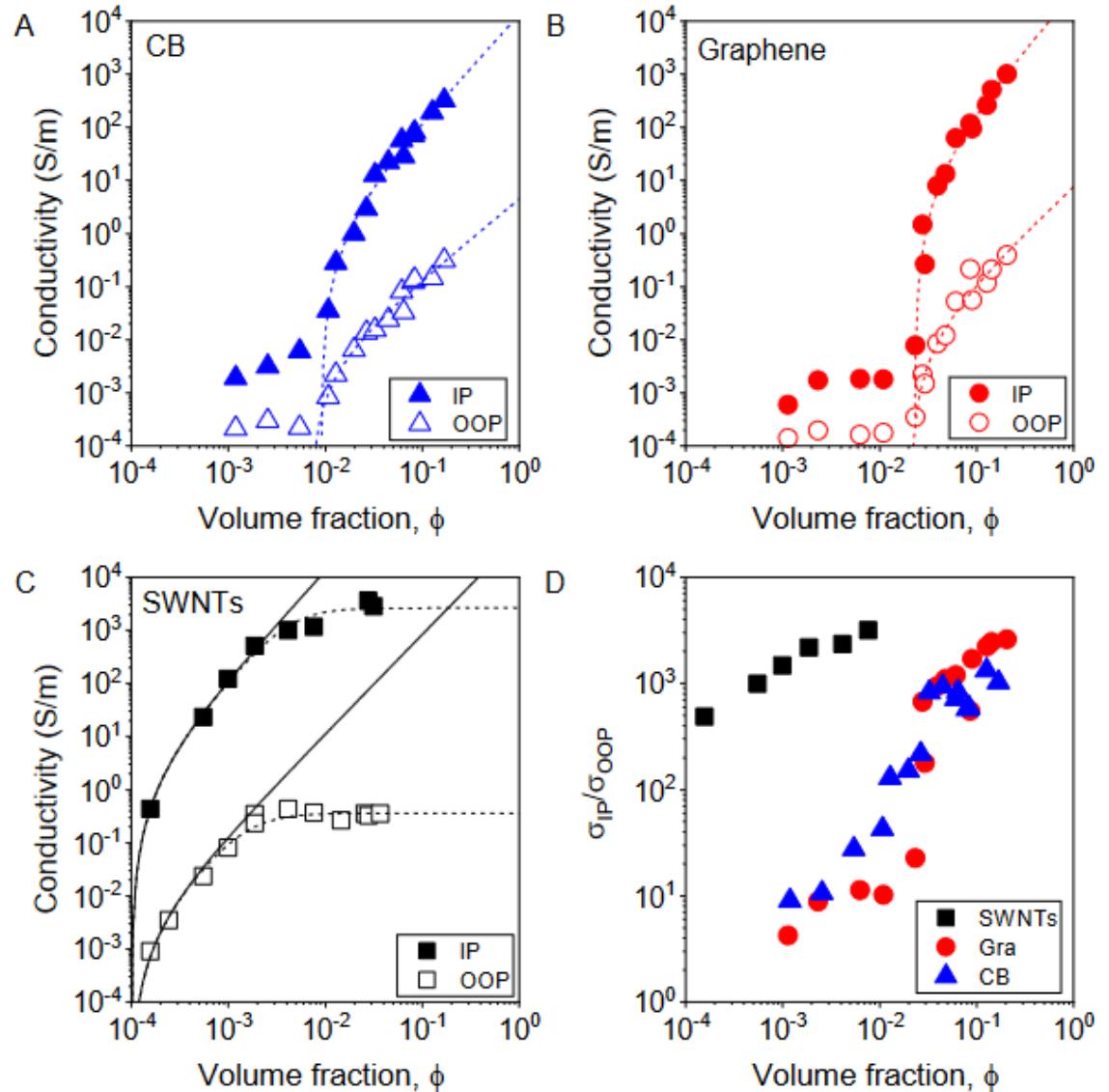
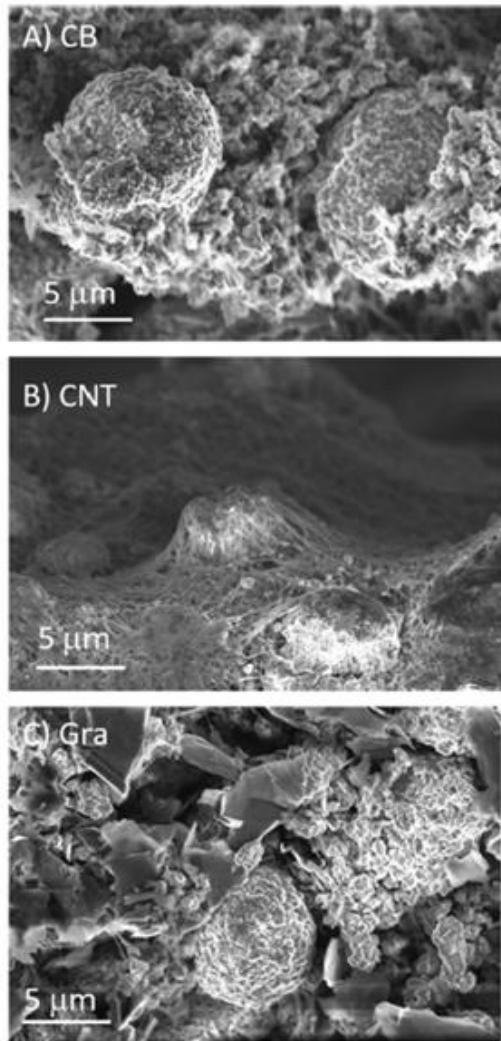
$$E_{s,am} \approx \frac{1}{2} * \frac{1}{2} * \frac{0.075 \text{ Ah}}{\text{kg}} * 4 \text{ V} = 75 \text{ Wh/kg}$$

Comparison of electrodes with different nanocarbons

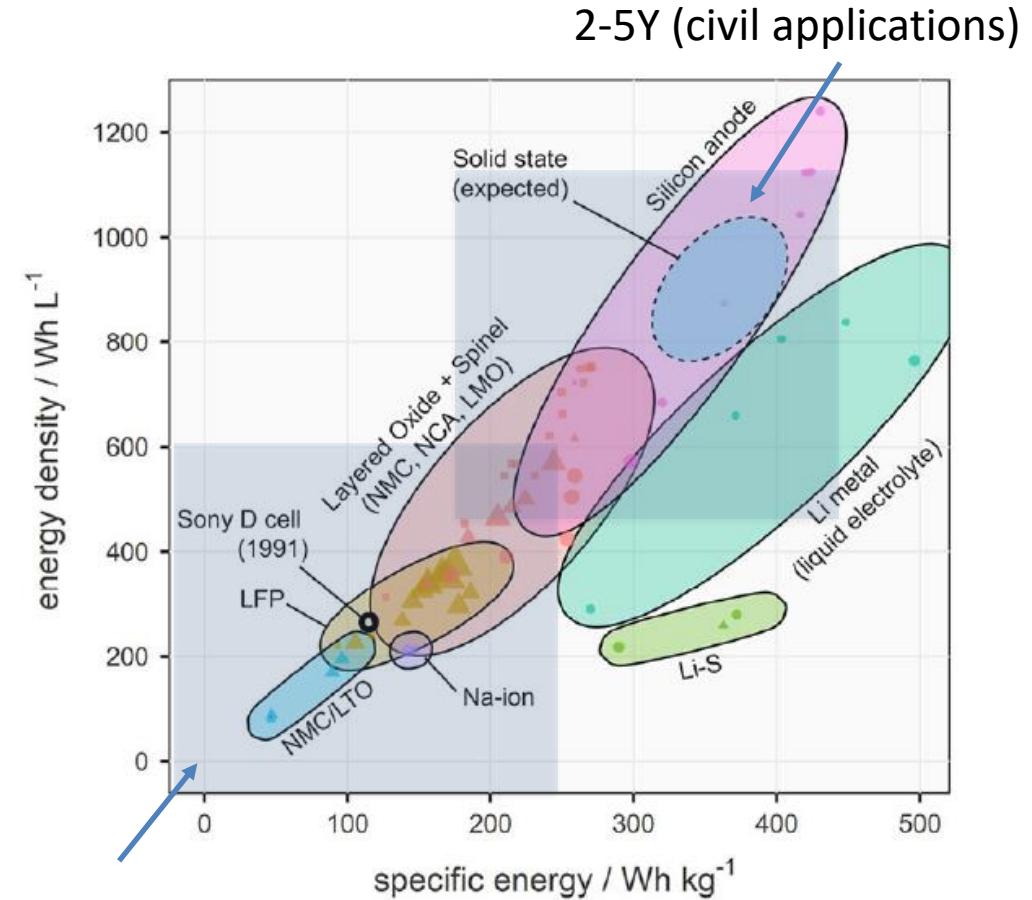
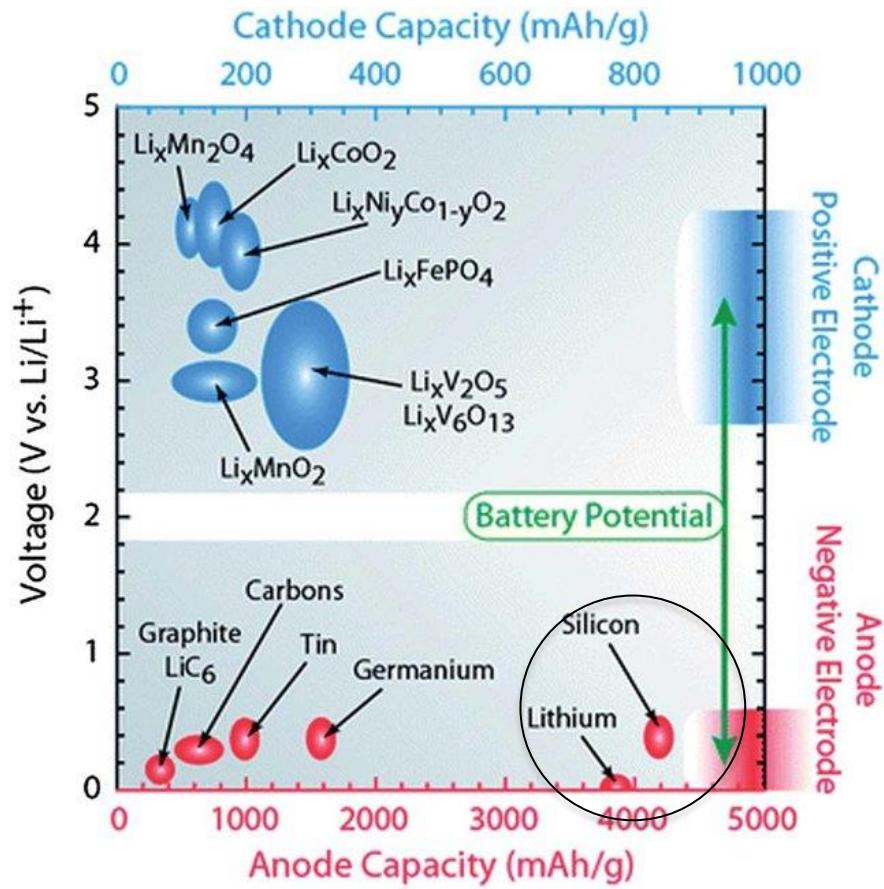


Electrodes of lithium–nickel–manganese–cobalt–oxide (NMC) with different nanocarbons and mass fractions

Comparison of electrodes of NMC with different nanocarbons

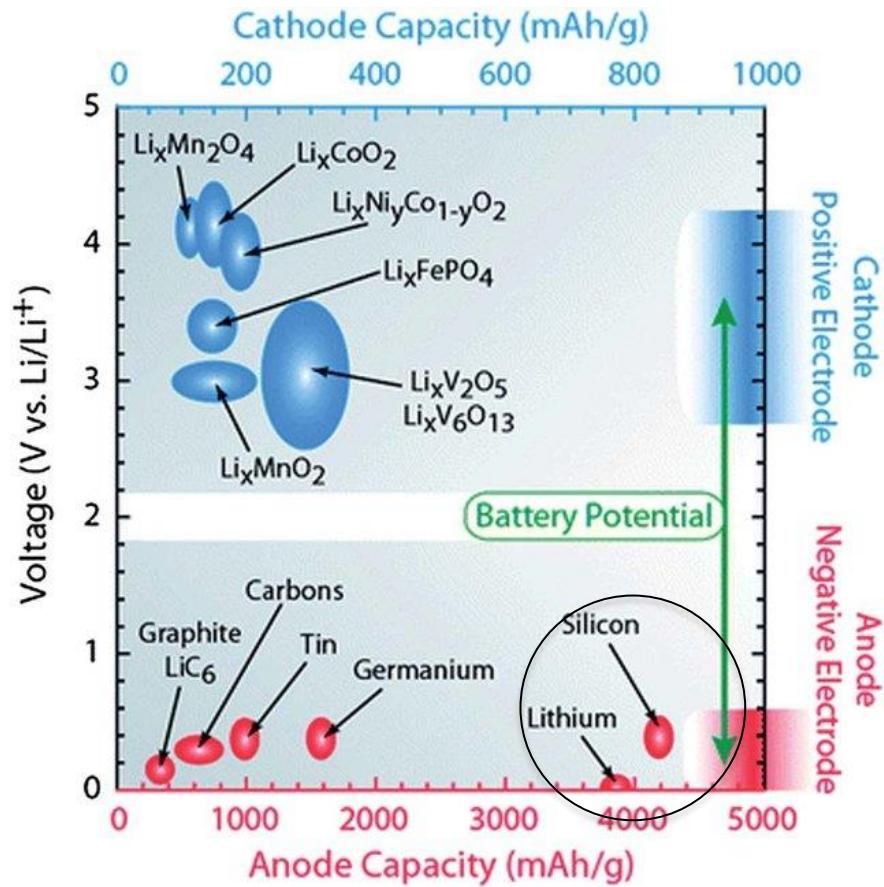


New generation of batteries



Today

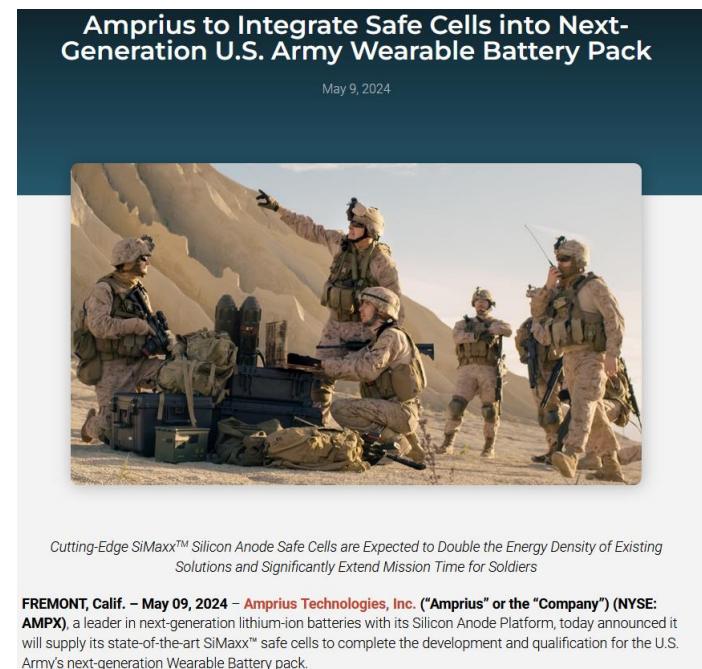
New generation of batteries



Energy density X 2 compared to legacy batteries (400 – 500 Wh/kg)

EV with 1000km drive range

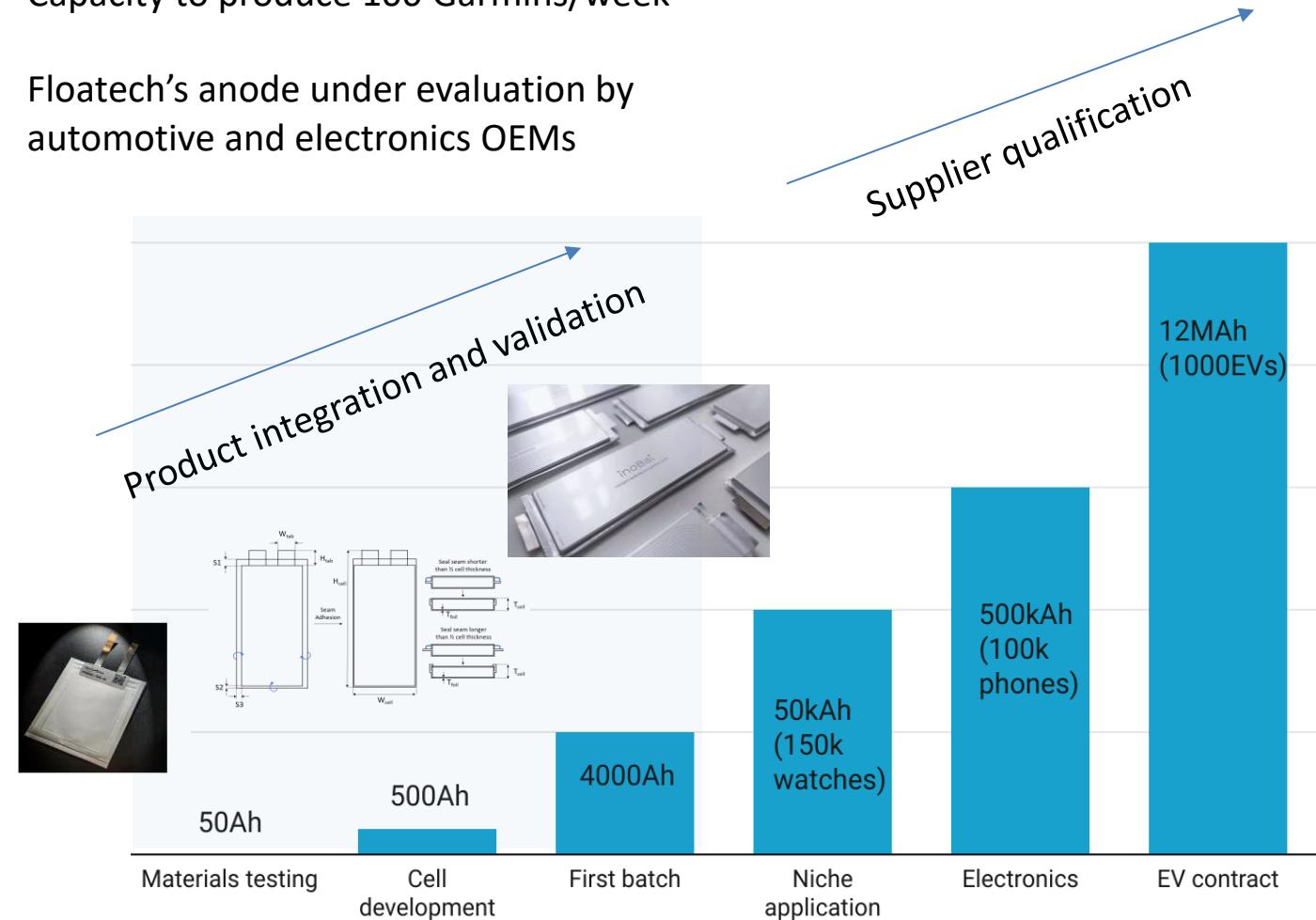
Charge time below 10s (10 – 80%)





Capacity to produce 100 Garmins/week

Floatech's anode under evaluation by automotive and electronics OEMs

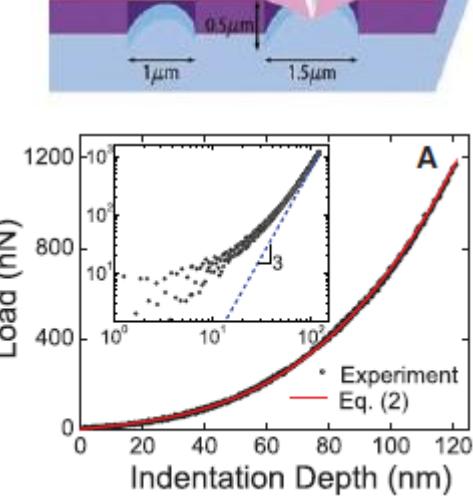
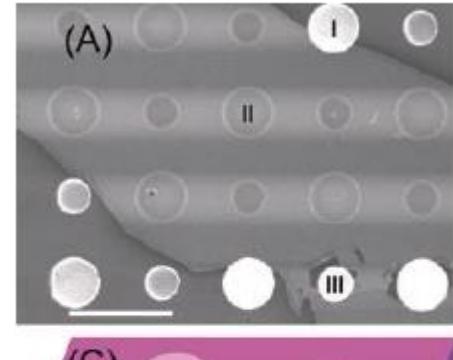


Cables eléctricos y escudos balísiticos nanoestructurados

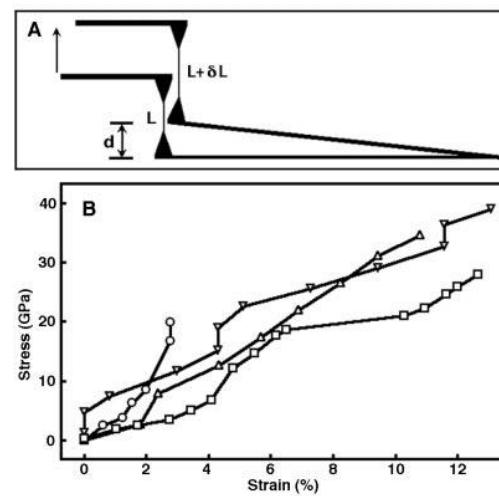
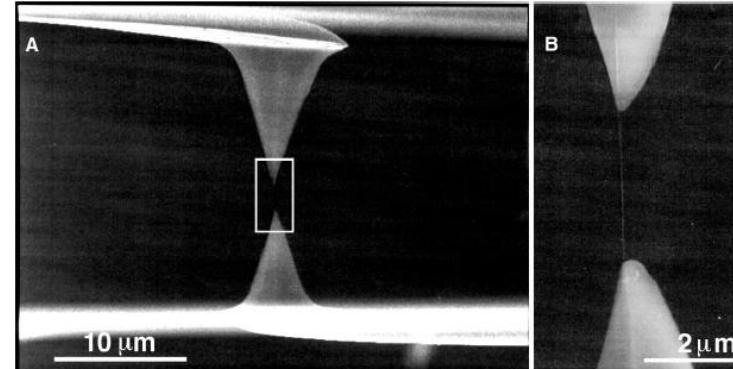
Motivation: properties of nanomaterials

Example: Graphene and CNTs

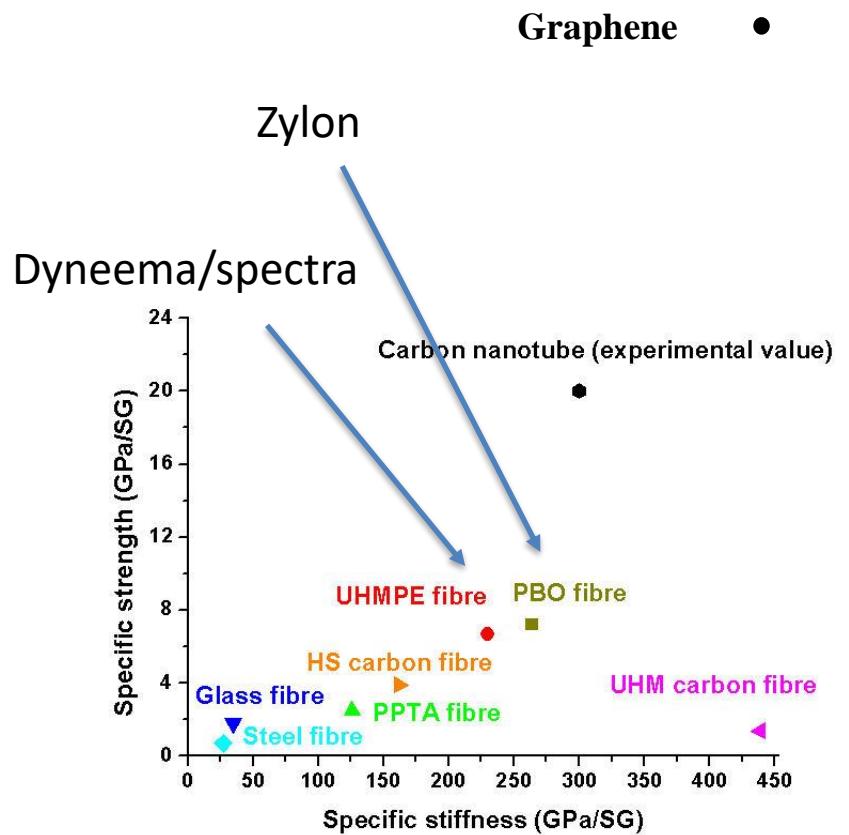
What is the origin of their outstanding mechanical properties?



Changgu Lee, et al. Science 321, 385 (2008)

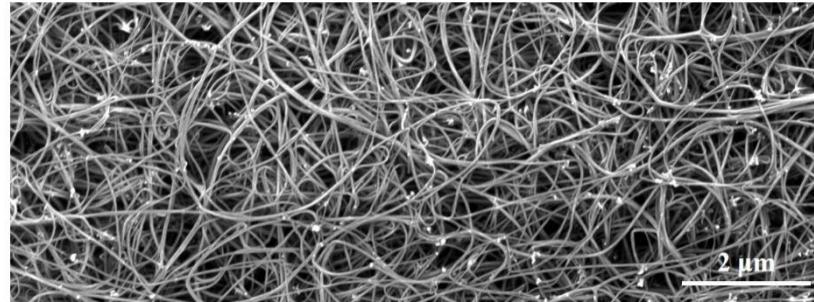
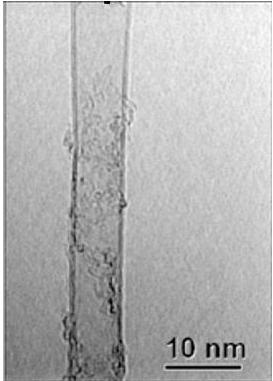


Yu et al. Science 287, 637 (2000)



“Nanomaterials” assembly into macroscopic materials

1D nanoparticle
(nanomer)

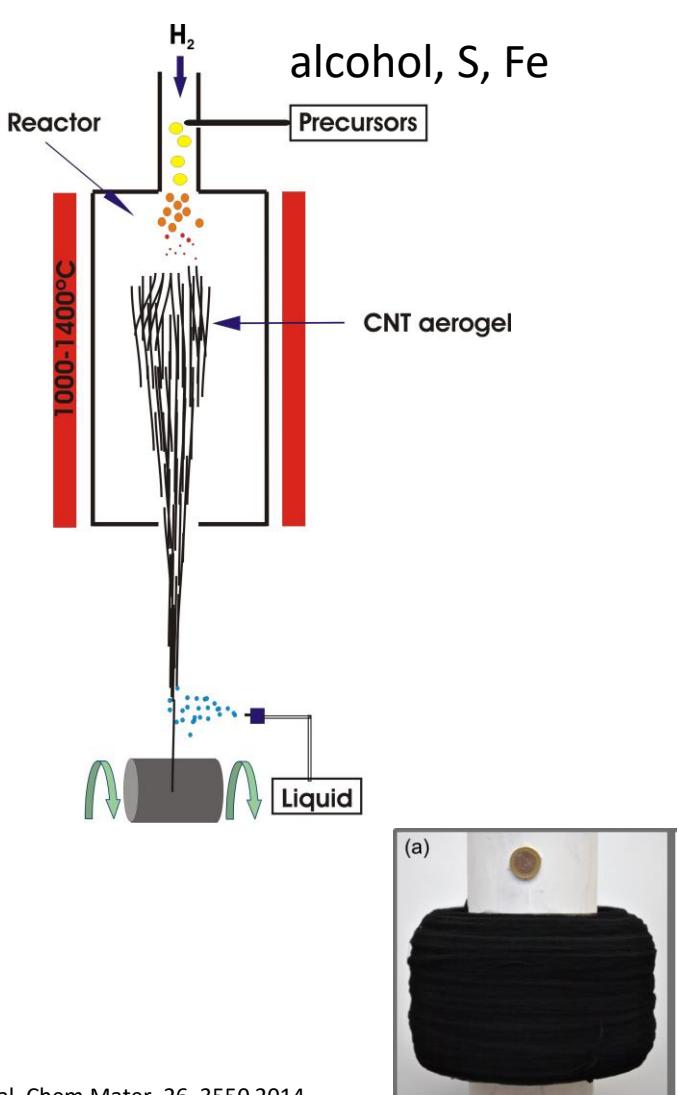


Engineering material
(nanometric material)



Macroscopic properties depend on assembly and microstructure

Synthesis by floating catalyst CVD



Reguero et al, Chem Mater, 26, 3550 2014

Looking up into the reactor

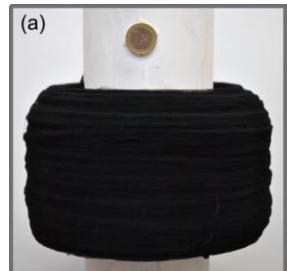


Continuous spinning of 1km



Synthesis by floating catalyst CVD

alcohol, S, Fe



Reguero et al, Chem Mater, 26, 3550 2014

Looking up into the reactor

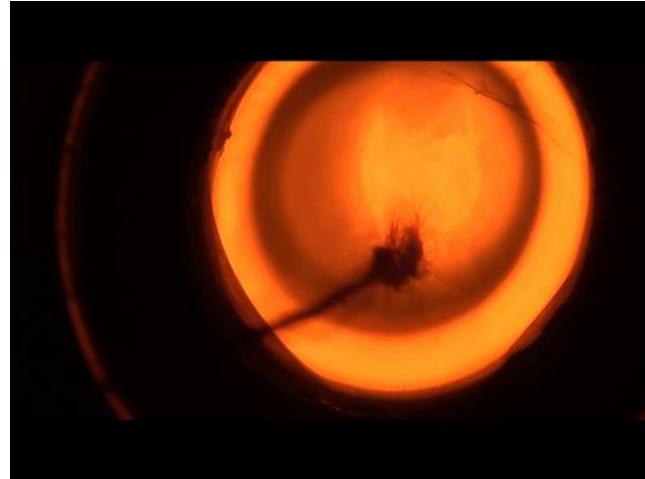


Continuous spinning of 1km

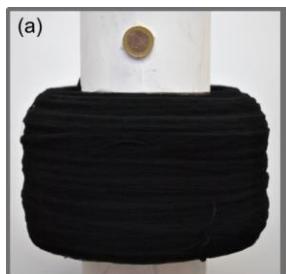


Synthesis by floating catalyst CVD

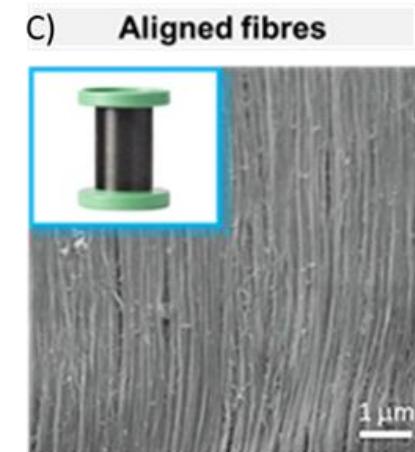
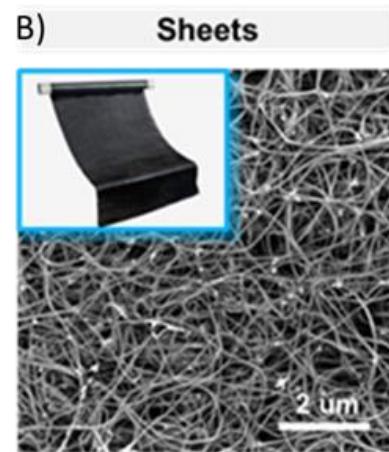
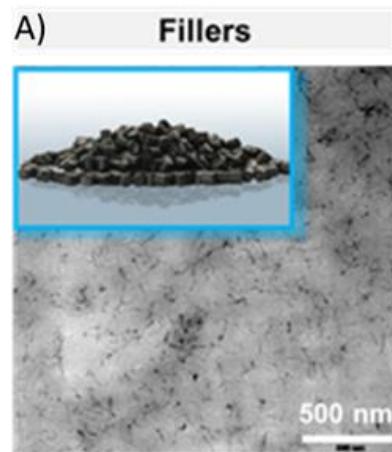
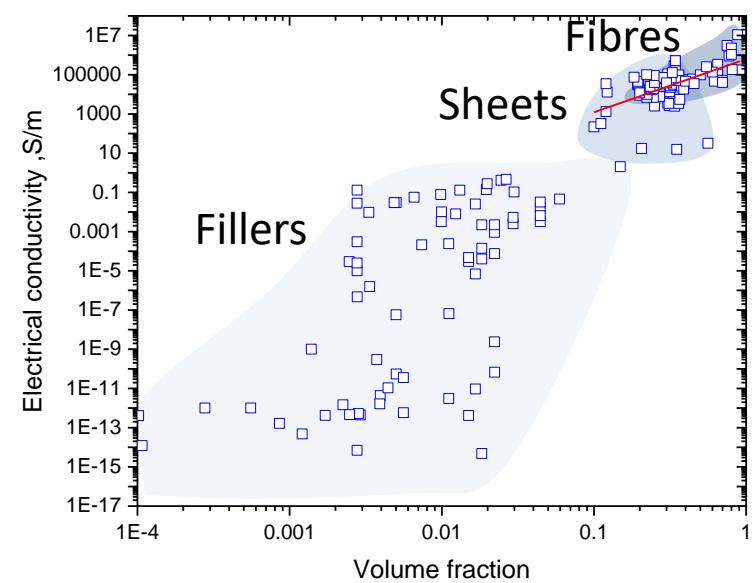
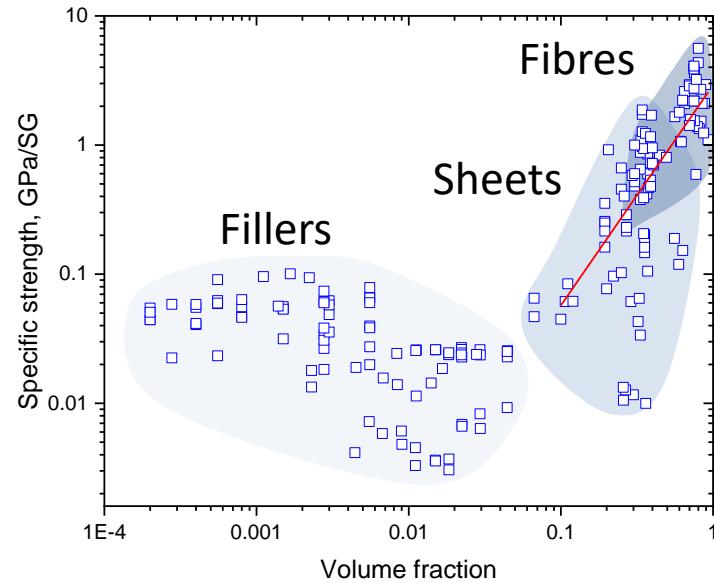
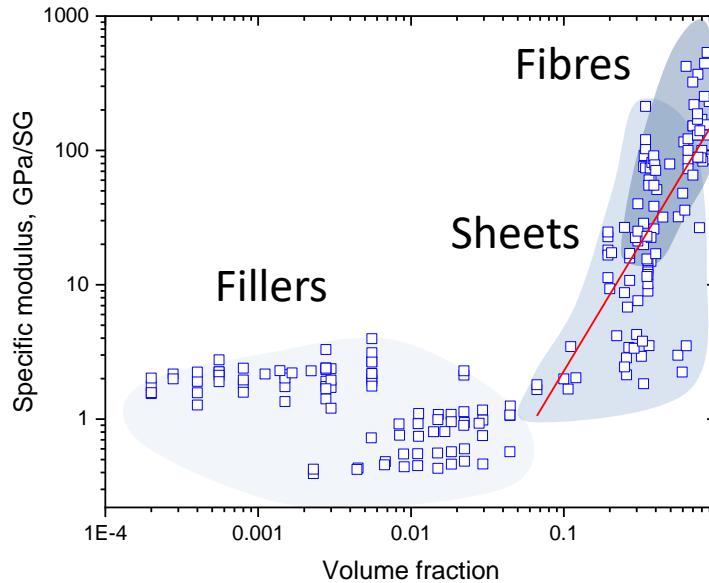
Looking up into the reactor



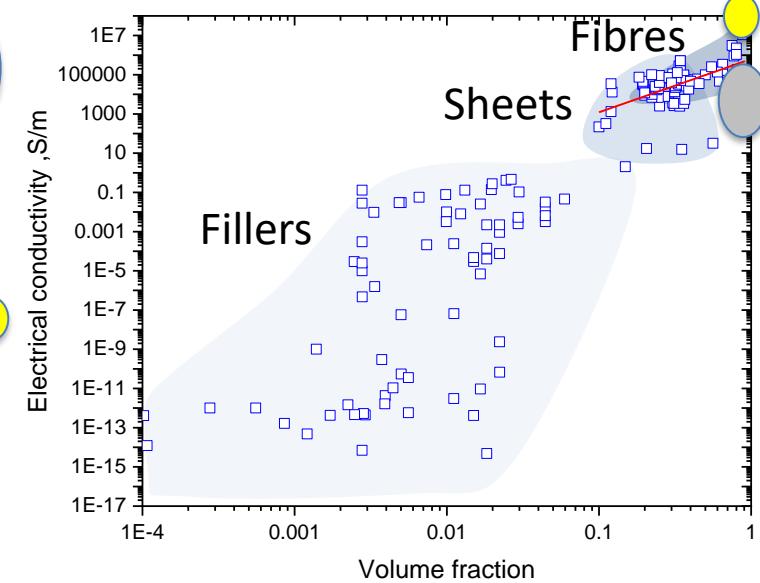
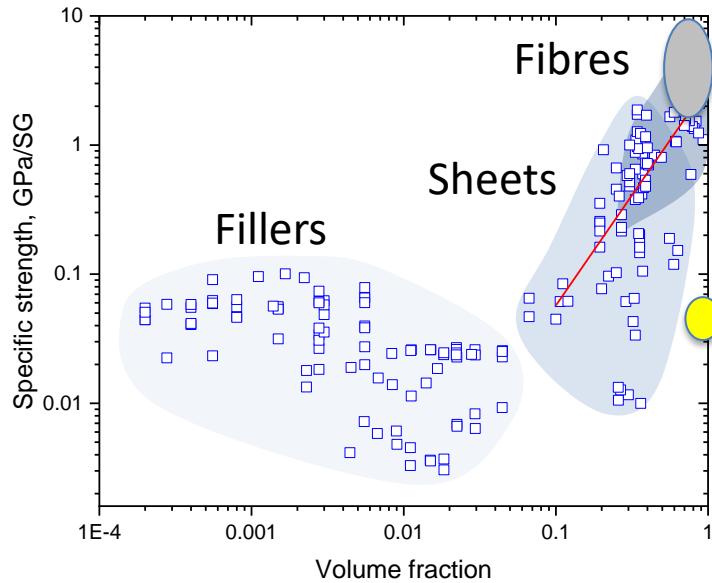
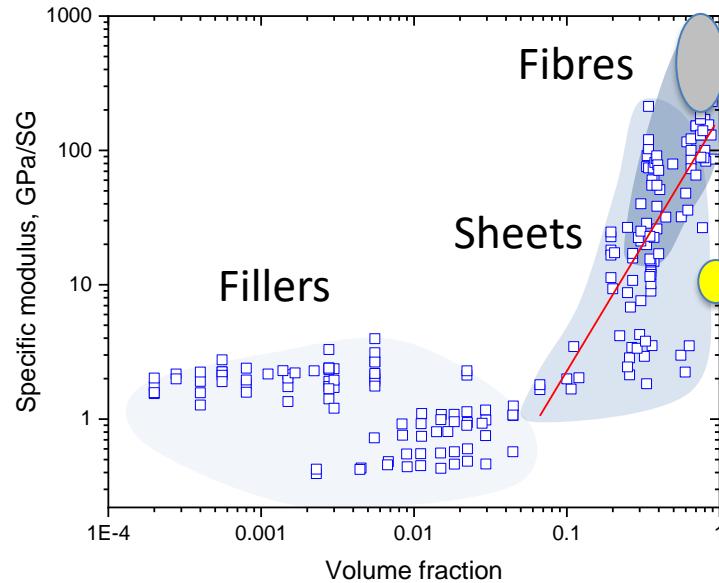
Continuous spinning of 1km



Classification of macromaterials of nanomaterials



Comparison to conventional materials



$$\text{Property} \propto V_f^\alpha$$

$\alpha = 1.9$ modulus; 1.7 strength, 2.7 conductivity

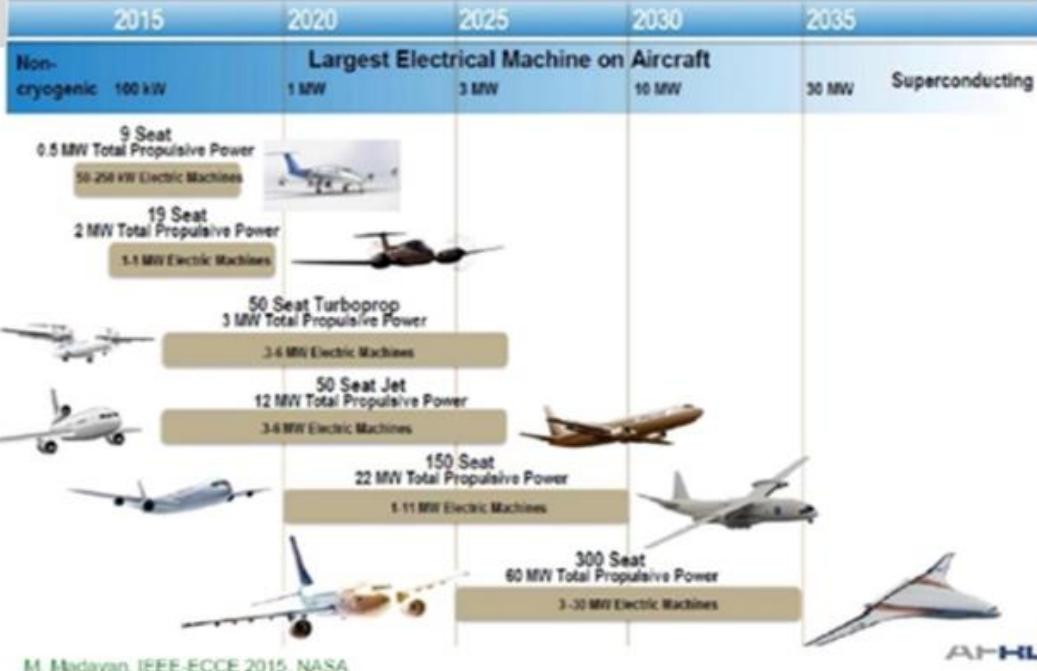
Carbon fibre
Copper

Value at scale, increases from 0.50 to 50 EUR/kg from fillers (CB) to fibres (CF)

Conductores supermetálicos: con conductividad y peso en el rango de Cu y Al

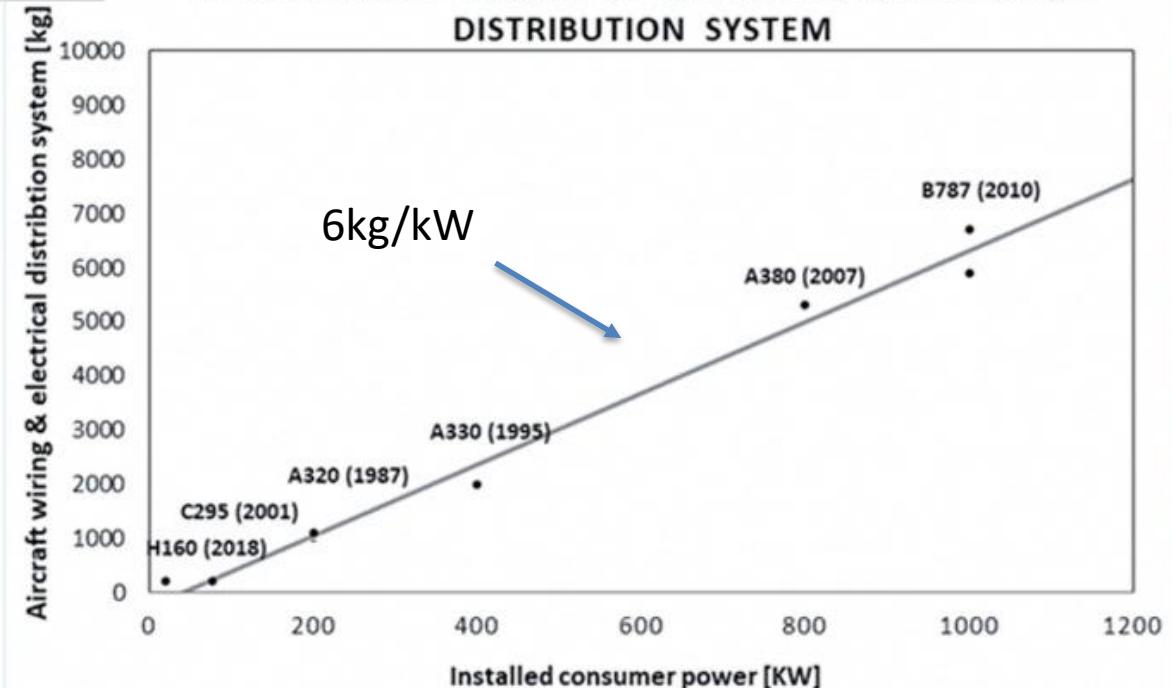
Motivation for new conductors for electrification of transport

a



b

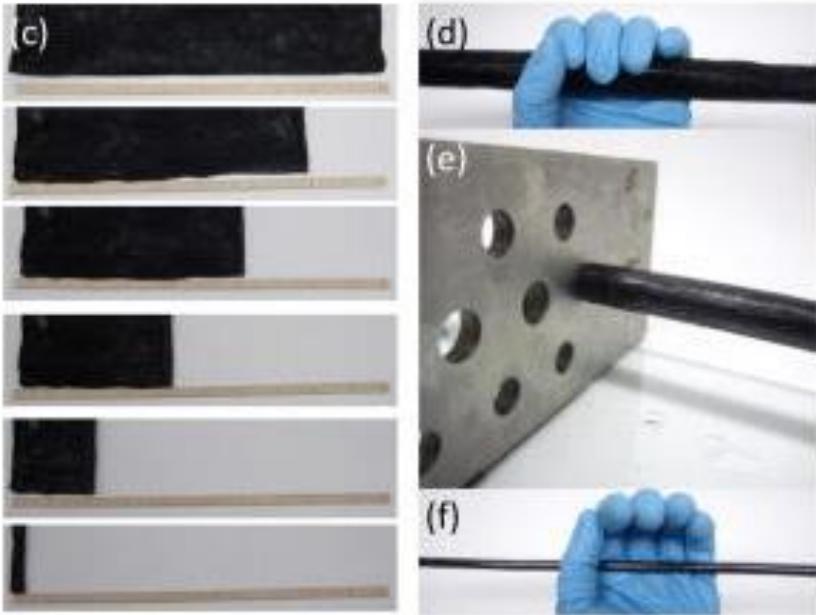
EVOLUTION OF WEIGHT OF AC WIRING & ELECTRICAL DISTRIBUTION SYSTEM



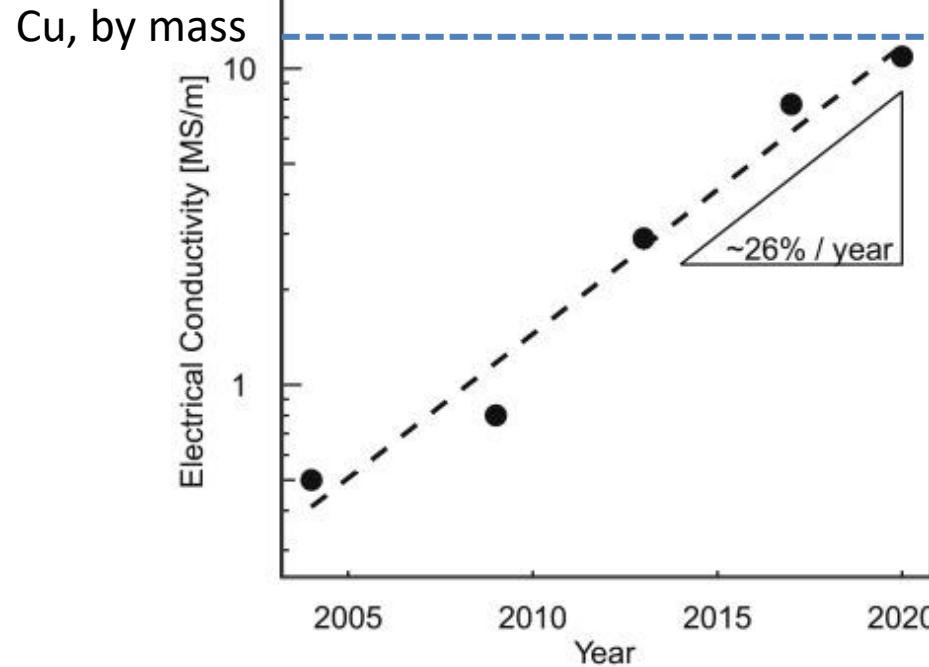
Increasing need for conductors in emerging aircraft. a) Evolution of required electrical power in civil aircraft [118] and b) Historical data showing a linear increase in weight from electrical wiring and distribution / interconnection systems with installed consumer power

At the current rate of increase in electrical power, by 2035 aircraft would carry wiring weighing as much as an A320!

Can we make cables lighter than copper?

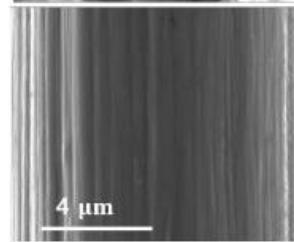
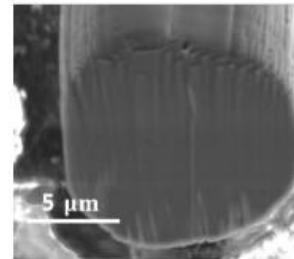
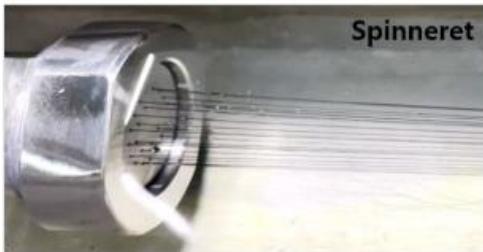


Cu

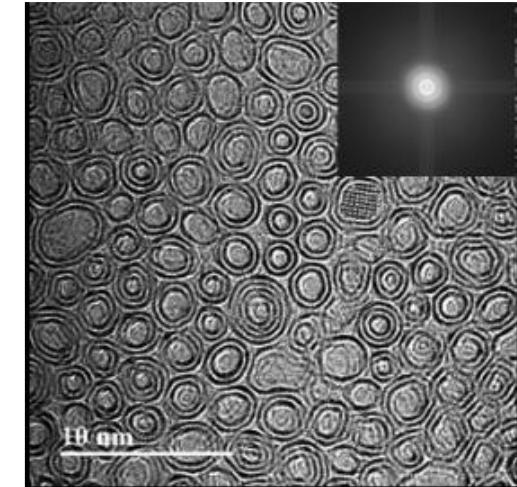


Cu: 7.5 MS/m /density
CNT: 4.8 MS/m /density

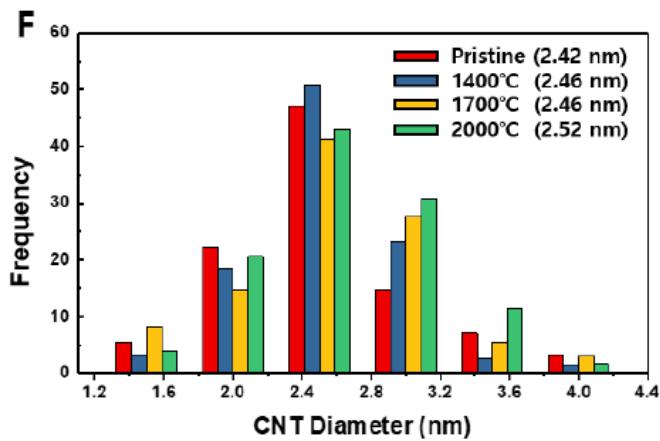
Properties of annealed LC-spun CNT fibres



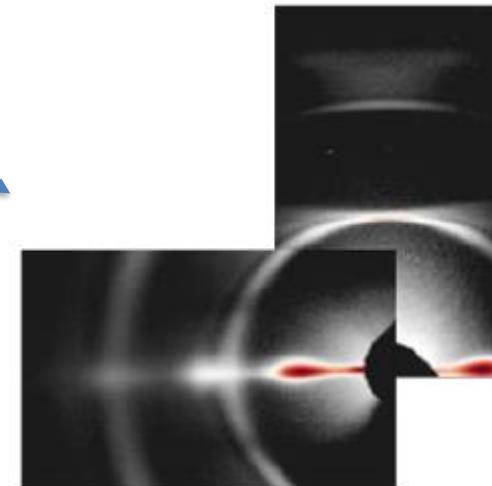
Periodic structure of closed-packed CNTs in regular bundles



Pt-protected FIB milling of lamellae



Long-range order

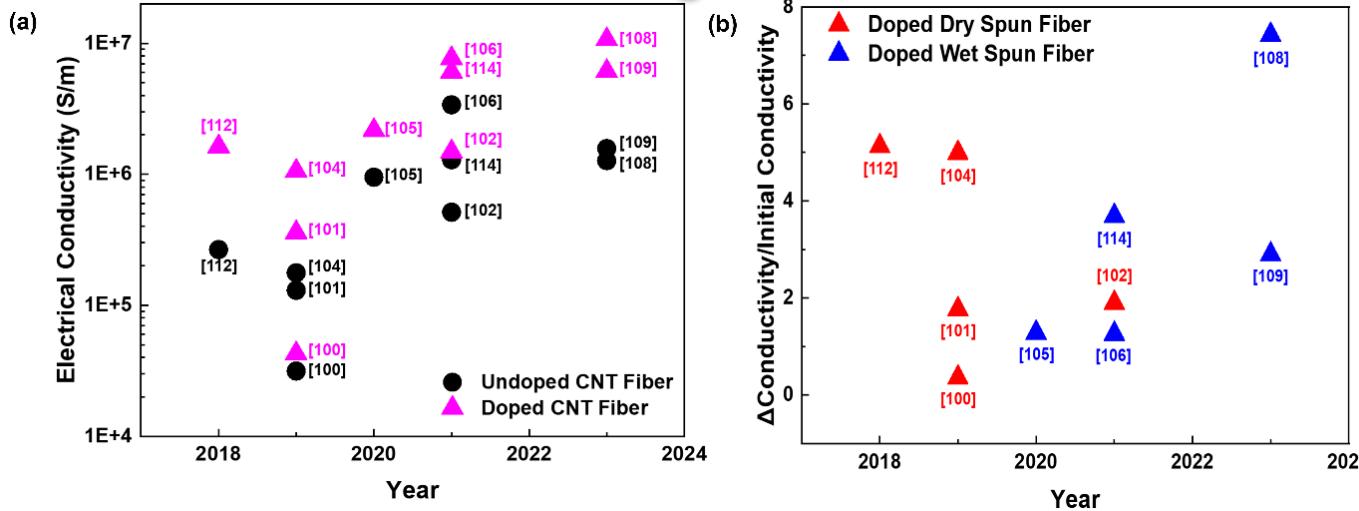


2D WAXS pattern

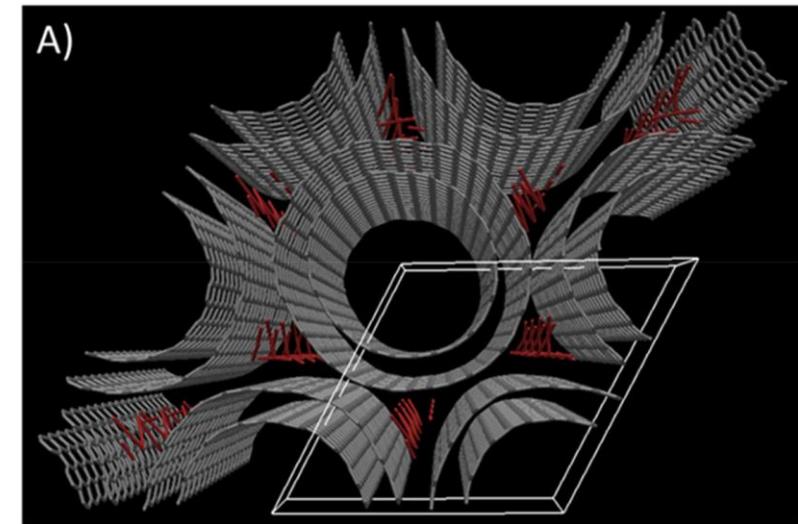
Can we make cables lighter than copper? Yes, we can



Upcoming IMDEA results



Conductivity of all CNT cables produced this decade

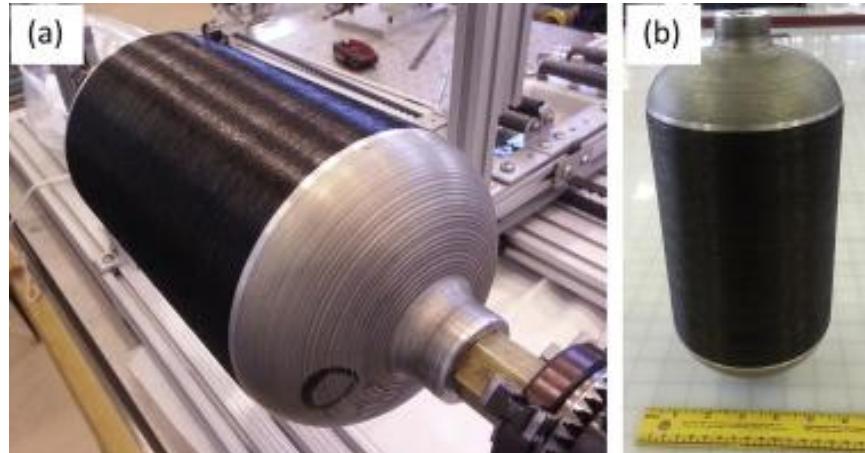


Cu: 7.5 MS/m /density

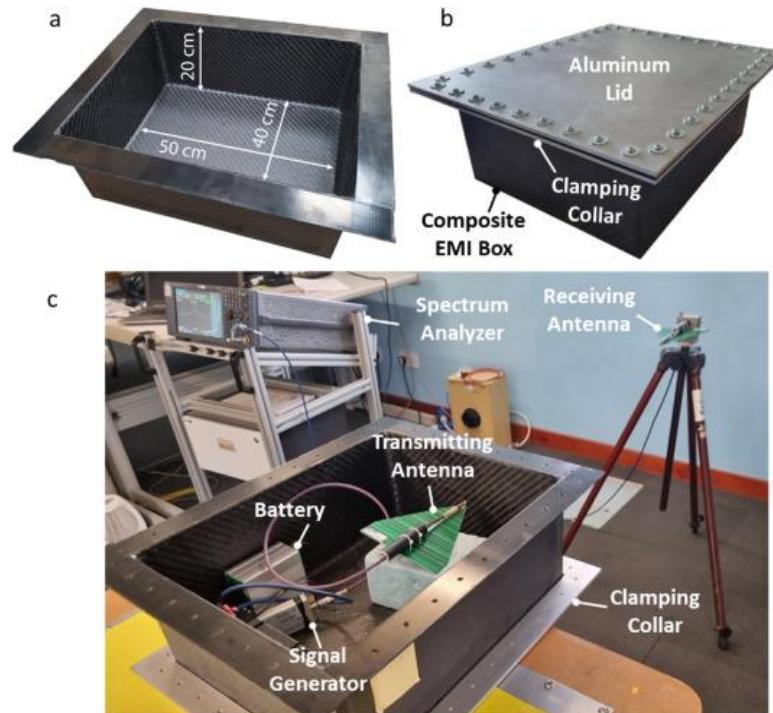
CNT: 4.8 MS/m /density (2024)
CNT: 8.3 MS/m / density (2025)

From lab to prototype

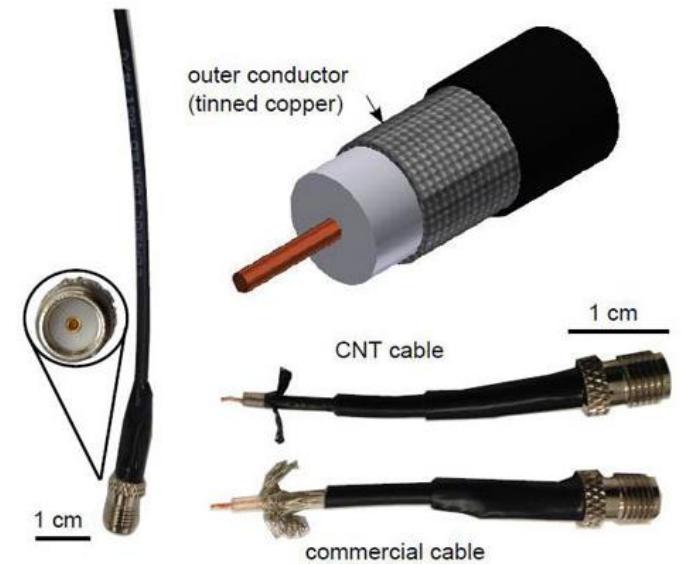
Nasa



Plasan, Israel



American Boronite Corporation



Perspective on weight reduction in electric vehicles



28% weight reduction in battery

25% weight reduction in wiring of electric motor

20-50% weight reduction in the EMI layer



Thank you

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Bon-Cheol Ku

Jun Yeon Hwang

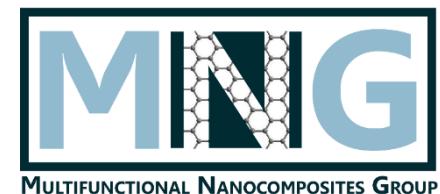
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(Texas A&M)

Adam Boies

(Cambridge)

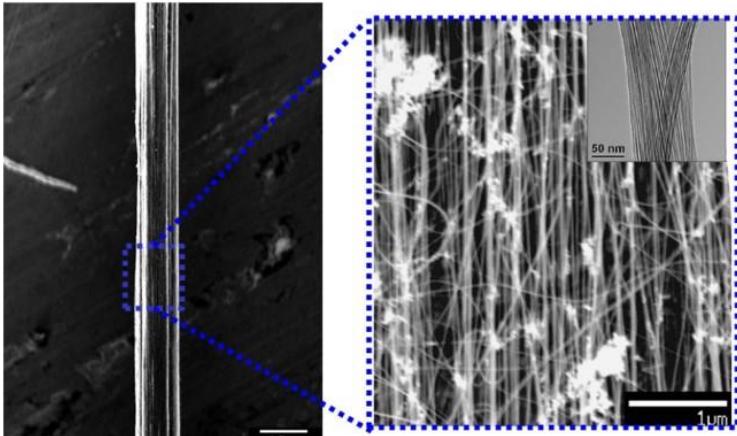


European
Research
Council

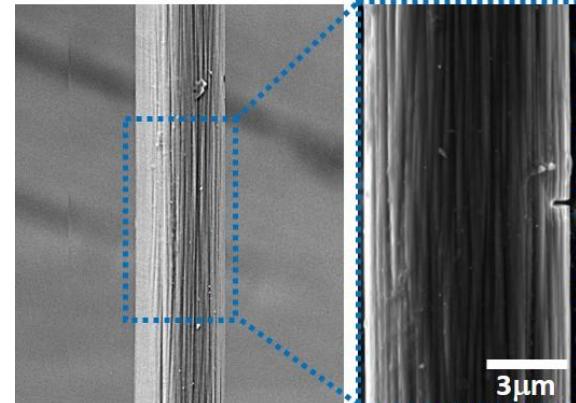


Tensile properties of aligned fibres and sheets

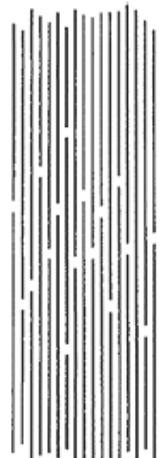
CNT Fibre



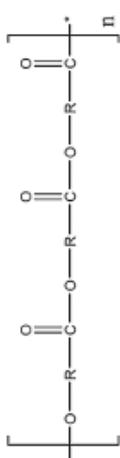
Carbon Fibre



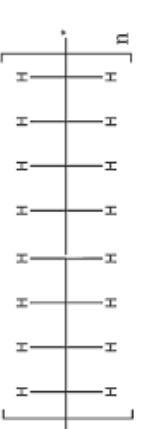
Staudinger model
1932



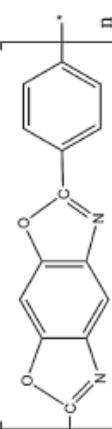
Polyester fibre
1932



UHMWPE fibre
early 70's



PBO fibre
late 70's



CNT fibre
2000

